FACULTY OF SCIENCE.

DEPARTMENT OF PHYSICS.

MAIN EXAMINATION 2010 / 2011

TITLE OF PAPER: SOLID STATE PHYSICS.

COURSE NUMBER: P 412.

TIME ALLOWED: THREE HOURS.

ANSWER ANY FOUR QUESTIONS . ALL CARRY EQUAL MARKS.

THIS PAPER IS NOT TO BE OPENED UNTIL PERMISSION HAS BEEN GIVEN BY THE INVIGILATOR.

Question One

- (a) Define the terms *lattice*, *primitive unit cell* and *conventional unit cell* of a crystal. (3 marks)
- (b) Draw *one* figure showing a conventional unit cell and also the primitive cell of an fcc lattice. (3 marks)
- (c) One side of a conventional unit cell of an f.c.c. lattice is 3 Å. What is the volume of its primitive unit cell? (2 marks)
- (d) Calculate the separation between two (123) planes of an orthorhombic lattice with cell lengths, a = 0.82 nm, b = 0.94nm and c = 0.75 nm in the x,y, z directions.

(4 marks)

- (e) (i) What is meant by packing fraction of a crystal? (2 marks)
 - (ii) Determine the packing fraction of a bcc crystal (4 marks)
- (f) A first order reflection from the (111) planes of a cubic crystal was observed at a glancing angle of 11.2° when x-rays of wavelength 154 pm were used. Calculate the length of the side of each cell. (1 pm = 10^{-12} m)

(3 marks)

(g) In the X-ray photograph of a cubic lattice, lines are observed at the following Bragg angles in degrees: 6.6, 9.2, 11.4, 13.1, 14.7. Identify the lattice type. (4 marks)

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- State what is meant by density of states in terms of frequency ω as applied in (i) (a) lattice dynamics.
 - By considering the boundary conditions on lattice waves in a cubic crystal, (ii) prove that for a three dimensional system, the density of states is given by the expression

$$D(\omega) = \frac{V\omega^2}{2\pi^2 v^3}$$

where V is the volume, v the velocity of sound.

(8 marks)

- What is Debye approximation? (3 marks) (b) (i)
 - (ii) Show that by applying Debye approximation, the density of states for a system with one atom per unit cell is given by

$$D(\omega) = \frac{9N\omega^2}{\omega_D^3}$$

where N is the number of atoms and ω_D , the Debye frequency. (7 marks)

Use the result in(b) above to show that the zero point energy of a lattice is given by (c)

$$E = \frac{9N}{8}\hbar\omega_D$$

[Given: the mean energy of a harmonic of oscillator is

$$\bar{\varepsilon} = \hbar\omega \left(\frac{1}{2} + \frac{1}{e^{\hbar\omega/kT} - 1}\right)$$
]. (5 marks)

Question Three

- (a) State the assumptions Drude made in his free electron theory of metals. (3 marks)
- (b) Define the terms *mean free path* and *mobility* of an electron. (2+2 marks)
- (c) Show that according to Drude theory the d.c electrical conductivity of a metal can be expressed as:

$$\sigma = \frac{ne^2 \tau}{m}$$
, where the symbols have their usual meanings. (10 marks)

- (d) (i) State Wiedemann Franz law. (3marks)
 - (ii) Write down the expression for *Lorenz number* and calculate its value. (2 + 3 marks)

Question Four

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- (a) (i) Define "Fermi energy". (2 marks)
 - (ii) Derive an expression for the density of states of a system of electrons, given that the Fermi energy:

$$\varepsilon_F = \frac{\hbar^2}{2m} \left(\frac{3\pi^2 N}{V} \right)^{2/3} \tag{6 marks}$$

- (iii) Calculate the density of energy states at 2.05 eV energy for a system of electrons in a volume of 1 cm³. (8 marks)
- (b) (i) Show that the electronic contribution to heat capacity of a metal is proportional to absolute temperature. (6 marks)
 - (ii) Discuss the heat capacity of metals, explaining the difference, if any, between the above theory and experimental values. (3 marks)

Question Five

(a) Using silicon as an example, explain how the electrical conductivity of a semiconductor can be increased by doping.

(6 marks)

(b) With the help of an appropriate diagram, derive an expression for the effective density of states in the conduction band of a semiconductor. Assume: $(\epsilon - \epsilon_F)$ » kT.

(10 marks)

Given:

Fermi -Dirac distribution function:
$$f(\varepsilon) = \frac{1}{e^{(\varepsilon - \varepsilon_F)/kT} + 1}$$

Density of states for a system of fermions:

$$D(\varepsilon)d\varepsilon = \frac{4\pi}{h^3}(2m)^{3/2}\,\varepsilon^{1/2}d\varepsilon$$

(c) A doped semiconductor has electron and hole concentrations $2x10^{13}$ cm⁻³ and 1.41×10^{13} cm⁻³ respectively. Calculate the electrical conductivity of the sample. (5 marks)

[Take:
$$\mu_n = 4200 \text{ cm}^2 \text{ V}^{\text{-1}} \text{s}^{\text{-1}}$$
. $\mu_p = ^*2000 \text{ cm}^2 \text{ V}^{\text{-1}} \text{s}^{\text{-1}}$]

(d) Discuss briefly the process of photoconductivity in semiconductors. (4 marks)

Various definite integrals.

$$\int_{0}^{\infty} e^{-ax^{2}} dx = \frac{1}{2} \sqrt{\frac{\pi}{a}}$$

$$\int_{0}^{\infty} e^{-ax^{2}} x dx = \frac{1}{2a}$$

$$\int_{0}^{\infty} e^{-ax^{2}} x^{3} dx = \frac{1}{2a^{2}}$$

$$\int_{0}^{\infty} e^{-ax^{2}} x^{2} dx = \frac{1}{4} \sqrt{\frac{\pi}{a^{3}}}$$

$$\int_{0}^{\infty} e^{-ax^{2}} x^{4} dx = \frac{3}{8a^{2}} \left(\frac{\pi}{a}\right)^{1/2}$$

$$\int_{0}^{\infty} e^{-ax^{2}} x^{5} dx = \frac{1}{a^{3}}$$

$$\int_{0}^{\infty} \frac{x^{3} dx}{e^{x} - 1} = \frac{\pi^{4}}{15}$$

$$\int_{0}^{\infty} x^{1/2} e^{-\lambda x} dx = \frac{\pi^{1/2}}{2\lambda^{3/2}}$$

$$\int_{0}^{\infty} \frac{x^{4}e^{x}}{(e^{x} - 1)^{2}} dx = \frac{4\pi^{4}}{15}$$

$$\int_{0}^{\infty} \frac{x^{1/2}}{e^{x} - 1} dx = \frac{2.61\pi^{1/2}}{2}$$

Appendix 2

Physical Constants.

Quantity	symbol	value
Speed of light Plank's constant	c h	3.00 x 10 ⁸ ms ⁻¹ 6.63 x 10 ⁻³⁴ J.s
Boltzmann constant	k	$1.38 \times 10^{-23} \text{ JK}^{-1}$
Electronic charge Mass of electron	e	1.61 x 10 ⁻¹⁹ C 9.11 x 10 ⁻³¹ kg
Mass of proton	m _e m _p	$1.67 \times 10^{-27 \text{kg}}$
Gas constant	R	8.31 J mol ⁻¹ K ⁻¹
Avogadro's number	N_A	6.02×10^{23}
Bohr magneton	$\mu_{\scriptscriptstyle \mathrm{B}}$	9.27 x 10 ⁻²⁴ JT ⁻¹ 4π x 10 ⁻⁷ Hm ⁻¹
Permeability of free space Stefan constant	$oldsymbol{\mu}_{ ext{o}}$ $oldsymbol{\sigma}$	$5.67 \times 10^{-8} \text{Wm}^{-2} \text{K}^{-4}$
Atmospheric pressure		$1.01 \times 10^5 \text{Nm}^{-2}$
Mass of ³ He atom		$6.65 \times 10^{-27} \text{ kg}$
Mass of 2 ³ He atom Volume of an ideal gas at STP		5.11 x 10 ⁻²⁷ kg 22.4 L mol ⁻¹