UNIVERSITY OF SWAZILAND

FACULTY OF SCIENCE

DEPARTMENT OF PHYSICS.

MAIN EXAMINATION 2006/2007

TITLE OF PAPER: STATISTICAL PHYSICS & THERMODYNAMICS

COURSE NUMBER: P 461

TIME ALLOWED: THREE HOURS

THIS PAPER CONTAINS FIVE QUESTIONS. ANSWER ANY  $\underline{FOUR}$  QUESTIONS . ALL QUESTIONS CARRY EQUAL MARKS

THIS PAPER IS NOT TO BE OPENED UNTIL PERMISSION HAS BEEN GIVEN BY THE INVIGILATOR.

### **OUESTION ONE.**

- (a) (i) Explain the difference between a **macrostate** and a **microstate** of a system of particles. (2 marks)
  - (ii) What is meant by <u>statistical weight</u> of a macrostate in a system of particles? (2 marks)
  - (iii) A classical system has 5 particles to be arranged in 3 energy levels.

    Using appropriate expression find the weights and hence the most probable distribution of these particles for the following cases:
  - 1. The energy levels are non-degenerate.
  - 2. The energy levels are doubly degenerate

(12 marks)

(b) (i) What is meant by Phase space?

(2 marks)

- (ii) Derive expressions for the volume element in phase space in terms of
  - 1. Momentum p
  - 2. Energy  $\epsilon$ .

(3 + 4 marks)

# **QUESTION TWO.**

(a) Derive the Fermi-Dirac distribution function for a system of fermions,

$$n_S = \frac{g_S}{e^{-(\alpha + \beta \epsilon_S)} + 1}$$
, where symbols have their usual meanings

(12 marks)

(b) (i) Given that the density of states for fermions is:

$$g(\varepsilon)d\varepsilon = \frac{4\pi V}{h^3}(2m)^{3/2}\varepsilon^{1/2}d\varepsilon$$

where symbols have their usual meanings, show that the Fermi energy of a system of fermions:

$$\varepsilon_F = \frac{h^2}{2m} \left(\frac{3N}{8\pi V}\right)^{2/3}$$

(8 marks)

(ii) Calculate the Fermi energy of a metal having density 8.5x10<sup>2</sup> kg m<sup>-3</sup> and atomic weight 40. (5 marks)

### **QUESTION THREE.**

(a) The Maxwell-Boltzmann distribution function for a system of classical particles is given by:

$$n_s = g_s e^{\alpha + \beta \varepsilon_s}$$
,

where the symbols have their usual meanings. Such a system has 2000 particles distributed in three non-degenerate energy levels having energies 1 unit and 2 units and 3 units each. The total energy is 2600 units. Use the above distribution function to obtain the values of  $\alpha$  and  $\beta$  of this system and hence find its probable configuration. Verify your answer numerically.

(15 marks)

(b) The differential form of Maxwell-Boltzmann distribution function in terms of the velocity v of the particles is given as:

$$n(v)dv = 4\pi N \left(\frac{m}{2\pi kT}\right)^{3/2} e^{-mv^2/2kT} v^2 dv$$

Use this expression to obtain:

- (i) The mean velocity
- (ii) The most probable velocity of the molecules of a classical gas.

[Note: see appendix for definite integrals ]

(10 marks)

## **QUESTION FOUR.**

(a) Use the Bose-Einstein distribution function of an assembly of identical non-interacting particles in thermal equilibrium to derive the Planck's radiation law for spectral distribution of energy radiated from a constant temperature enclosure.

(9 marks)

- (b) Obtain an expression for the total energy per unit volume emitted from the enclosure at temperature T. (5 marks)
- (c) (i) State briefly what is **Bose\_Einstein condensation**. (3 marks)
  - (ii) The density of ideal gas consisting of particles having mass  $6.65 \times 10^{-27}$  kg is  $1.17 \times 10^{26}$  m<sup>-3</sup>.
    - 1. Calculate the Bose temperature  $T_B$  of the gas. (5 marks)
    - 2. What fraction of the particles will be in the ground state at a temperature of  $0.1T_{\rm B}$ . (3 marks)

Given: 
$$N = 2.612V \left(\frac{2\pi mkT_B}{h^2}\right)^{3/2}$$

# **QUESTION FIVE.**

(a) Derive the partition function of a classical gas:

$$Z = \frac{v}{h^3} (2\pi mkT)^{3/2}$$

(8 marks)

(b) Show that the pressure of the classical gas:

$$P = NkT \frac{\partial \ln Z}{\partial V}$$

Hence derive the ideal gas equation P V = N k T

(10 marks)

(c) Calculate the translational partition function of an hydrogen molecule confined to a volume of  $100~{\rm cm}^3$  at  $300~{\rm K}$ .

(7 marks)

## Appendix 1

Various definite integrals.

$$\int_{0}^{\infty} e^{-ax^{2}} dx = \frac{1}{2} \sqrt{\frac{\pi}{a}}$$

$$\int_{0}^{\infty} e^{-ax^{2}} x dx = \frac{1}{2a}$$

$$\int_{0}^{\infty} e^{-ax^{2}} x^{3} dx = \frac{1}{2a^{2}}$$

$$\int_{0}^{\infty} e^{-ax^{2}} x^{2} dx = \frac{1}{4} \sqrt{\frac{\pi}{a^{3}}}$$

$$\int_{0}^{\infty} e^{-ax^{2}} x^{4} dx = \frac{3}{8a^{2}} \left(\frac{\pi}{a}\right)^{1/2}$$

$$\int_{0}^{\infty} e^{-ax^{2}} x^{5} dx = \frac{1}{a^{3}}$$

$$\int_{0}^{\infty} \frac{x^{3} dx}{e^{x} - 1} = \frac{\pi^{4}}{15}$$

$$\int_{0}^{\infty} x^{1/2} e^{-\lambda x} dx = \frac{\pi^{1/2}}{2\lambda^{3/2}}$$

$$\int_{0}^{\infty} \frac{x^{4}e^{x}}{(e^{x} - 1)^{2}} dx = \frac{4\pi^{4}}{15}$$

$$\int_{0}^{\infty} \frac{x^{1/2}}{e^{x} - 1} dx = \frac{2.61\pi^{1/2}}{2}$$

# Appendix 2

Physical Constants.

Quantity	symbol	value
Speed of light	С	3.00 x 10 <sup>8</sup> ms <sup>-1</sup>
Plank's constant	h	6.63 x 10 <sup>-34</sup> J.s
Boltzmann constant	$\mathbf{k}^{\cdot}$	1.38 x 10 <sup>-23</sup> JK <sup>-1</sup>
Electronic charge	е	1.61 x 10 <sup>-19</sup> C
Mass of electron	$m_e$	9.11 x 10 <sup>-31</sup> kg
Mass of proton	$m_{p}$	$1.67 \times 10^{-27 \text{ kg}}$
Gas constant	R	8.31 J mol <sup>-1</sup> K <sup>-1</sup>
Avogadro's number	$N_A$	$6.02 \times 10^{23}$
Bohr magneton	$\mu_{\scriptscriptstyle  m B}$	9.27 x 10 <sup>-24</sup> JT <sup>-1</sup>
Permeability of free space	$\mu_{o}$	$4\pi \times 10^{-7} \text{Hm}^{-1}$
Stefan constant	σ	$5.67 \times 10^{-8} \text{Wm}^{-2} \text{K}^{-4}$
Atmospheric pressure		1.01 x 10 <sup>5</sup> Nm <sup>-2</sup>
Mass of 24 He atom		6.65 x 10 <sup>-27</sup> kg
Mass of 2 <sup>3</sup> He atom		5.11 x 10 <sup>-27</sup> kg
Volume of an ideal gas at STP		22.4 l mol <sup>-1</sup>