

UNIVERSITY OF SWAZILAND
FACULTY OF SCIENCE

DEPARTMENT OF PHYSICS

SUPPLEMENTARY EXAMINATION 2005

TITLE OF THE PAPER: NUCLEAR PHYSICS

COURSE NUMBER: P442

TIME ALLOWED: THREE HOURS

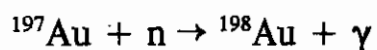
INSTRUCTIONS: CHOOSE AND ATTEMPT ANY FOUR QUESTIONS.
EACH QUESTION CARRIES 25 MARKS.

THE PAPER CONSISTS OF SEVEN PAGES INCLUDING THIS ONE. DATA SHEETS
ARE ALSO INCLUDED.

THIS PAPER IS NOT TO BE OPENED UNTIL PERMISSION HAS BEEN GIVEN BY
THE INVIGILATOR

Question 1

- a) A sample of gold is exposed to a neutron beam of constant intensity such that 10^{10} neutrons per second are absorbed in the reaction



The nuclide ^{198}Au undergoes β^- - decay to ^{198}Hg with a mean life of 3.89 days.

- (i) Show the decay scheme. (3 marks)
- (ii) How many atoms of ^{198}Au will be present after six days of irradiation? (11 marks)
- (iii) How many atoms of ^{198}Hg will be present at that time, assuming that the ^{198}Hg is unaffected by the neutron beam? (6 marks)
- (iv) What is the equilibrium number of ^{198}Au atoms? (5 marks)

Question 2

- a) What is the mechanical potential energy of an electron at the centre of a gold nucleus ($Z=79$, $A=197$) given that the nucleus is a uniformly charged sphere of radius $R=1.2 A^{1/3}$ fm? Consider the cases when the observer is outside and inside the sphere. (15 marks)
- b) Estimate the radius of the first Bohr orbit of a μ - meson about
- (i) a proton (5 marks)
 - (ii) a ^{12}C nucleus (5 marks)

Question 3

- a) Explain the terms in the semi-empirical mass formula

$$M(Z, A) = ZM_H + NM_n - a_v A + a_s A^{2/3} + a_c \frac{Z^2}{A^{1/3}} + a_{\text{asym}} \frac{(A-2Z)^2}{A} \pm a/A^{1/3} \quad (15 \text{ marks})$$

- b) Show for large A and Z that the energy released when a nucleus (Z,A) emits an α -particle is given by

$$Q = -4a_v + 8a_s/3A^{1/3} + 4a_c Z(1-Z/3A)/A^{1/3} - 4a_{\text{asym}}(N-Z)^2/A^2 + B(2,4)$$

where B (2,4) is the binding energy of the α -particle, 28.30 MeV.

(10 marks)

Question 4

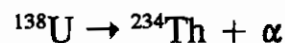
- a) (i) Explain why naturally occurring nuclei of mass number $A < 60$ are not α -unstable. (4 marks)

- (ii) With neglect of nuclear spins, and for Coulomb potential barriers large compared with the α -particle energy, the dependence of the meanlife τ of an α -unstable nucleus on the charge Z and α -particle velocity v is represented by

$$\tau \propto \exp(4\pi Ze^2/4\pi\epsilon_0\hbar v)$$

Discuss the basic assumptions made in the derivation of this expression. (4 marks)

- b) In the decay

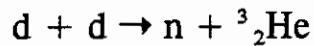


the emission of α -particles of energy 4.195, 4.147, 4.038 MeV is observed. Use the formula in (a) (ii) above to estimate the relative strength of these lines.

- (i) What are the most probable spin-parity assignments for the states of ^{234}Th produced in this reaction? (9 marks)
- (ii) State briefly why consideration of the spins of the states of ^{234}Th would change your estimates of the relative line strengths. (8 marks)

Question 5

- a) Compute the Q-values for the reactions



given mass defect of the neutron = 0.008 665 u
 mass defect of deuterium atom = 0.014 102 u
 mass defect of tritium atom = 0.016 050 u
 mass defect of helium - 3 atom = 0.016 030 u
 mass defect of helium - 4 atom = 0.002 603 u

(8 marks)

- b) What are the maximum energies (in the laboratory) of neutrons that can be produced using 4 MeV deuterons incident on stationary targets of deuterium and of tritium? Use non-relativistic kinematics and assume nuclear masses are

$$A \times 931.5 \text{ MeV} / c^2$$

in the transformation, but use the Q-values from (a) above. (17 marks)

Question 6

- a) (i) Assuming that the wavefunction

$$u(r) = r\psi(r) = Ce^{-\alpha r}$$

is valid for the deuteron from $r=0$ to $r=\infty$ obtain the value of the normalization constant C .

- (ii) If $\alpha=0.232\text{fm}^{-1}$, find the probability that the separation of the two particles in the deuteron exceeds a value of 2fm

- (iii) Find the average distance of interaction for this wavefunction.

- b) (i) For the two-body system the radial Schrodinger wave equation is given as

$$\frac{d^2}{dr^2}u_\ell + \frac{2m}{\hbar^2}\left[E - V - \frac{\ell(\ell+1)\hbar^2}{2mr^2}\right]u_\ell = 0 \quad 6.1$$

where ℓ is orbital angular momentum. The last term in the square bracket is called the centrifugal potential. Sketch the behaviour of this potential and state its importance.

- (ii) To simplify the derivation it is usual to assume the s-state for the deuteron wave function. Discuss this assumption in relation to the fact that the observed deuteron spin is 1.

CONSTANTS

Speed of light	c	$2.99792458 \times 10^8 \text{ m/s}$
Charge of electron	e	$1.602189 \times 10^{-19} \text{ C}$
Boltzmann constant	k	$1.38066 \times 10^{-23} \text{ J/K}$
		$8.6174 \times 10^{-5} \text{ eV/K}$
Planck's constant	h	$6.62618 \times 10^{-34} \text{ J} \cdot \text{s}$
		$4.13570 \times 10^{-15} \text{ eV} \cdot \text{s}$
	$\hbar = h/2\pi$	$1.054589 \times 10^{-34} \text{ J} \cdot \text{s}$
		$6.58217 \times 10^{-16} \text{ eV} \cdot \text{s}$
Gravitational constant	G	$6.6726 \times 10^{-11} \text{ N} \cdot \text{m}^2/\text{kg}^2$
Avogadro's number	N_A	$6.022045 \times 10^{23} \text{ mole}^{-1}$
Universal gas constant	R	$8.3144 \text{ J/mole} \cdot \text{K}$
Stefan-Boltzmann constant	σ	$5.6703 \times 10^{-8} \text{ W/m}^2 \cdot \text{K}^4$
Rydberg constant	R_∞	$1.0973732 \times 10^7 \text{ m}^{-1}$
Hydrogen ionization energy		13.60580 eV
Bohr radius	a_0	$5.291771 \times 10^{-11} \text{ m}$
Bohr magneton	μ_B	$9.27408 \times 10^{-24} \text{ J/T}$
		$5.78838 \times 10^{-5} \text{ eV/T}$
Nuclear magneton	μ_N	$5.05084 \times 10^{-27} \text{ J/T}$
		$3.15245 \times 10^{-8} \text{ eV/T}$
Fine structure constant	α	$1/137.0360$
	$\hbar c$	$1239.853 \text{ MeV} \cdot \text{fm}$
	$\hbar c$	$197.329 \text{ MeV} \cdot \text{fm}$
	$e^2/4\pi\epsilon_0$	$1.439976 \text{ MeV} \cdot \text{fm}$

PARTICLE REST MASSES

	u	MeV/c^2
Electron	5.485803×10^{-4}	0.511003
Proton	1.00727647	938.280
Neutron	1.00866501	939.573
Deuteron	2.01355321	1875.628
Alpha	4.00150618	3727.409
$\pi^\pm$	0.1498300	139.5669
π^0	0.1448999	134.9745
μ	0.1134292	105.6595

CONVERSION FACTORS

$$1 \text{ eV} = 1.602189 \times 10^{-19} \text{ J}$$

$$1 u = 931.502 \text{ MeV}/c^2$$

$$= 1.660566 \times 10^{-27} \text{ kg}$$

$$1 \text{ b} = 10^{-28} \text{ m}^2$$

$$1 \text{ Ci} = 3.7 \times 10^{10} \text{ decays/s}$$

PERIODIC TABLE OF THE ELEMENTS

PERIODIC TABLE OF THE ELEMENTS																						
1 H 1.0079	3 Li 6.941	4 Be 9.012																2 He 4.0026				
11 Na 22.990	12 Mg 24.305																5 B 10.81	6 C 12.011	7 N 14.007	8 O 15.999	9 F 18.998	10 Ne 20.179
19 K 39.098	20 Ca 40.08	21 Sc 44.956	22 Ti 47.90	23 V 50.942	24 Cr 51.996	25 Mn 54.938	26 Fe 55.847	27 Co 58.933	28 Ni 58.70	29 Cu 63.546	30 Zn 65.38	31 Ga 69.72	32 Ge 72.59	33 As 74.922	34 Se 78.96	35 Br 79.904	36 Kr 83.80	54 Xe 131.30				
37 Rb 85.468	38 Sr 87.62	39 Y 88.906	40 Zr 91.22	41 Nb 92.906	42 Mo 95.94	43 Tc (97)	44 Ru 101.07	45 Rh 102.91	46 Pd 106.4	47 Ag 107.87	48 Cd 112.41	49 In 114.82	50 Sn 118.69	51 Sb 121.75	52 Te 127.60	53 I 126.90	54 Xe 131.30	86 Rn (222)				
55 Cs 132.91	56 Ba 137.33	57-71 Rare Earths	72 Hf 178.49	73 Ta 180.95	74 W 183.85	75 Re 186.21	76 Os 190.2	77 Ir 192.22	78 Pt 195.09	79 Au 196.97	80 Hg 200.59	81 Tl 204.37	82 Pb 207.2	83 Bi 208.98	84 Po (209)	85 ^{At} (210)	86 Rn (222)					
87 Fr (223)	88 Ra (226)	89-103 Acti- nides	104 Rf (257)	105 Ha (263)	106 (263)	107	108	109														
Rare Earths (Lanthanides)																						
57 La 138.91	58 Ce 140.12	59 Pr 140.91	60 Nd 144.24	61 Pm (145)	62 Sm 150.4	63 Eu 151.96	64 Gd 157.25	65 Tb 158.93	66 Dy 162.50	67 Ho 164.93	68 Er 167.26	69 Tm 168.93	70 Yb 173.04	71 Lu 174.97								
89 Ac (227)	90 Th 232.04	91 Pa (231)	92 U 238.03	93 Np (237)	94 Pu (244)	95 Am (243)	96 Cm (247)	97 Bk (247)	98 Cf (251)	99 Es (254)	100 Fm (257)	101 Md (258)	102 No (259)	103 Lr (260)								
Actinides																						

For each element, the bottom line gives either the atomic mass or (in parentheses, for radioactive elements) the mass number of the most stable isotope.

TABLE OF NUCLEAR PROPERTIES

The following table shows some properties of a selection of isotopes. For each element only the stable and relatively long-lived radioactive isotopes are included. Ground-state atomic masses and spin-parity assignments are shown for all isotopes; uncertain spin-parity assignments are in parentheses. Abundances are given for stable isotopes, and for radioactive isotopes the half-life and principal decay mode are shown (ϵ —electron capture, possibly including positron emission; β^- —negative beta decay; α —alpha decay; f—spontaneous fission). The masses are those of the corresponding neutral atoms and were taken from the 1983 atomic mass evaluation: A. H. Wapstra and G. Audi, *Nucl. Phys. A* 432, 1 (1985). In the half-life entries, My = 10^6 y. Uncertainties in the masses are typically 10^{-5} u (10^{-4} u for some cases far from stability); uncertainties in the abundances and half-lives are typically at or below the level of the last digit tabulated.

					Abundance or Half-life						Abundance or Half-life
Z	A	Atomic mass (u)	I^π			Z	A	Atomic mass (u)	I^π		
H	1	1.007825	$\frac{1}{2}^+$		99.985%	10	10.012937	3^+			19.8%
	2	2.014102	1^+		0.015%	11	11.009305	$\frac{3}{2}^-$			80.2%
	3	3.016049	$\frac{1}{2}^+$		12.3 y (β^-)	12	12.014353	1^+			20.4 ms (β^-)
He	2	3.016029	$\frac{1}{2}^+$		$1.38 \times 10^{-4}\%$	13	13.017780	$\frac{3}{2}^-$			17.4 ms (β^-)
	4	4.002603	0^+		99.99986%	C	6	9.031039	$\frac{3}{2}^-$		0.13 s (ϵ)
Li	3	6.015121	1^+		7.5%		10	10.016856	0^+		19.2 s (ϵ)
	7	7.016003	$\frac{3}{2}^-$		92.5%		11	11.011433	$\frac{3}{2}^-$		20.4 m (ϵ)
	8	8.022486	2^+		0.84 s (β^-)		12	12.000000	0^+		98.89%
Be	4	7.016928	$\frac{3}{2}^-$		53.3 d (ϵ)		13	13.003355	$\frac{1}{2}^-$		1.11%
	8	8.005305	0^+		0.07 fs (α)		14	14.003242	0^+		5730 y (β^-)
	9	9.012182	$\frac{3}{2}^-$		100%		15	15.010599	$\frac{1}{2}^+$		2.45 s (β^-)
	10	10.013534	0^+		1.6 My (β^-)	N	7	12.018613	1^+		11 ms (ϵ)
	11	11.021658	$\frac{1}{2}^+$		13.8 s (β^-)		13	13.005739	$\frac{1}{2}^-$		9.96 m (ϵ)
B	5	8.024606	2^+		0.77 s (ϵ)		14	14.003074	1^+		99.63%
	9	9.013329	$\frac{3}{2}^-$		0.85 as (α)		15	15.000109	$\frac{1}{2}^-$		0.366%
							16	16.006100	2^-		7.13 s (β^-)

Abundance or Half-life					Abundance or Half-life						
Z	A	Atomic mass (u)	I ^π	Half-life	Z	A	Atomic mass (u)	I ^π	Half-life		
O	8	17	17.008450	$\frac{1}{2}^{-}$	4.17 s (β ⁻)	26	25.986892	5 ⁺	0.72 My (ε)		
		18	18.014081	1 ⁻	0.63 s (β ⁻)	27	26.981539	$\frac{5}{2}^{+}$	100 %		
		14	14.008595	0 ⁺	71 s (ε)	28	27.981910	3 ⁺	2.24 m (β ⁻)		
		15	15.003065	$\frac{1}{2}^{-}$	122 s (ε)	29	28.980446	$\frac{5}{2}^{+}$	6.6 m (β ⁻)		
		16	15.994915	0 ⁺	99.76 %	30	29.982940	3 ⁺	3.7 s (β ⁻)		
		17	16.999131	$\frac{5}{2}^{+}$	0.038 %	Si	14	25.992330	0 ⁺	2.21 s (ε)	
		18	17.999160	0 ⁺	0.204 %		27	26.986704	$\frac{5}{2}^{+}$	4.13 s (ε)	
		19	19.003577	$\frac{5}{2}^{+}$	26.9 s (β ⁻)		28	27.976927	0 ⁺	92.23 %	
20	20.004076	0 ⁺	13.5 s (β ⁻)	29	28.976495		$\frac{1}{2}^{+}$	4.67 %			
F	9	17	17.002095	$\frac{5}{2}^{+}$	64.5 s (ε)		30	29.973770	0 ⁺	3.10 %	
		18	18.000937	1 ⁺	110 m (ε)		31	30.975362	$\frac{3}{2}^{+}$	2.62 h (β ⁻)	
		19	18.998403	$\frac{1}{2}^{+}$	100 %		32	31.974148	0 ⁺	105 y (β ⁻)	
		20	19.999981	2 ⁺	11 s (β ⁻)		33	32.997920	($\frac{1}{2}^{+}$)	6.2 s (β ⁻)	
		21	20.999948	$\frac{5}{2}^{+}$	4.3 s (β ⁻)	P	15	28.981803	$\frac{1}{2}^{+}$	4.1 s (ε)	
		22	22.003030	(3, 4) ⁺	4.2 s (β ⁻)		30	29.978307	1 ⁺	2.50 m (ε)	
		23	23.003600	($\frac{3}{2}, \frac{5}{2}$) ⁺	2.2 s (β ⁻)		31	30.973762	$\frac{1}{2}^{+}$	100 %	
		Ne	10	17	17.017690		$\frac{1}{2}^{-}$	0.11 s (ε)	32	31.973907	1 ⁺
18	18.005710			0 ⁺	1.7 s (ε)		33	32.971725	$\frac{1}{2}^{+}$	25.3 d (β ⁻)	
19	19.001880			$\frac{1}{2}^{+}$	17.3 s (ε)		34	33.973636	1 ⁺	12.4 s (β ⁻)	
20	19.992436			0 ⁺	90.51 %	S	16	30	29.984903	0 ⁺	1.2 s (ε)
21	20.993843			$\frac{3}{2}^{+}$	0.27 %		31	30.979554	$\frac{1}{2}^{+}$	2.6 s (ε)	
22	21.991383			0 ⁺	9.22 %		32	31.972071	0 ⁺	95.02 %	
23	22.994465			$\frac{5}{2}^{+}$	37.6 s (β ⁻)		33	32.971458	$\frac{3}{2}^{+}$	0.75 %	
24	23.993613			0 ⁺	3.4 m (β ⁻)		34	33.967867	0 ⁺	4.21 %	
25	24.997690			($\frac{1}{2}, \frac{3}{2}$) ⁺	0.60 s (β ⁻)		35	34.969032	$\frac{3}{2}^{+}$	87.4 d (β ⁻)	
Na	11	20	20.007344	2 ⁺	0.45 s (ε)		36	35.967081	0 ⁺	0.017 %	
		21	20.997651	$\frac{3}{2}^{+}$	22.5 s (ε)		37	36.971126	$\frac{7}{2}^{-}$	5.0 m (β ⁻)	
		22	21.994434	3 ⁺	2.60 y (ε)		38	37.971162	0 ⁺	170 m (β ⁻)	
		23	22.989768	$\frac{3}{2}^{+}$	100 %	Cl	17	33	32.977452	$\frac{3}{2}^{+}$	2.51 s (ε)
		24	23.990961	4 ⁺	15.0 h (β ⁻)		34	33.973763	0 ⁺	1.53 s (ε)	
		25	24.989953	$\frac{5}{2}^{+}$	60 s (β ⁻)		35	34.968853	$\frac{3}{2}^{+}$	75.77 %	
		26	25.992586	3 ⁺	1.1 s (β ⁻)		36	35.968307	2 ⁺	0.30 My (β ⁻)	
		27	26.993940	$\frac{5}{2}^{+}$	0.30 s (β ⁻)		37	36.965903	$\frac{1}{2}^{+}$	24.23 %	
Mg	12	21	21.011716	($\frac{3}{2}, \frac{5}{2}$) ⁺	0.123 s (ε)		38	37.968011	2 ⁻	37.3 m (β ⁻)	
		22	21.999574	0 ⁺	3.86 s (ε)		39	38.968005	$\frac{3}{2}^{+}$	56 m (β ⁻)	
		23	22.994124	$\frac{3}{2}^{+}$	11.3 s (ε)		40	39.970440	2 ⁻	1.35 m (β ⁻)	
		24	23.985042	0 ⁺	78.99 %		41	40.970590	($\frac{1}{2}, \frac{3}{2}$) ⁺	31 s (β ⁻)	
		25	24.985837	$\frac{5}{2}^{+}$	10.00 %	Ar	18	34	33.980269	0 ⁺	0.844 s (ε)
		26	25.982594	0 ⁺	11.01 %		35	34.975256	$\frac{1}{2}^{+}$	1.78 s (ε)	
		27	26.984341	$\frac{1}{2}^{+}$	9.46 m (β ⁻)		36	35.967546	0 ⁺	0.337 %	
		28	27.983877	0 ⁺	21.0 h (β ⁻)		37	36.966776	$\frac{3}{2}^{+}$	35.0 d (ε)	
Al	13	28	28.988480	$\frac{3}{2}^{+}$	1.4 s (β ⁻)		38	37.962732	0 ⁺	0.063 %	
		24	23.999941	4 ⁺	2.07 s (ε)		39	38.964314	$\frac{7}{2}^{-}$	269 y (β ⁻)	
		25	24.990429	$\frac{5}{2}^{+}$	7.18 s (ε)		40	39.962384	0 ⁺	99.60 %	
							41	40.964501	$\frac{7}{2}^{-}$	1.83 h (β ⁻)	

				Abundance or Half-life					Abundance or Half-life		
Z	A	Atomic mass (u)	I ^π		Z	A	Atomic mass (u)	I ^π			
	42	41.963050	0 ⁺	33 y (β ⁻)		52	51.946898	0 ⁺	1.7 m (β ⁻)		
	43	42.965670		5.4 m (β ⁻)		53	52.949730	($\frac{3}{2}$) ⁻	33 s (β ⁻)		
	44	43.965365	0 ⁺	11.9 m (β ⁻)							
K	19	37	36.973377	$\frac{3}{2}^{+}$	V	23	46	45.960198	0 ⁺	0.42 s (ε)	
		38	37.969080	3 ⁺			47	46.954906	$\frac{1}{2}^{-}$	32.6 m (ε)	
		39	38.963707	$\frac{1}{2}^{+}$			48	47.952257	4 ⁺	16.0 d (ε)	
		40	39.963999	4 ⁻			49	48.948517	$\frac{7}{2}^{-}$	330 d (ε)	
		41	40.961825	$\frac{1}{2}^{+}$			50	49.947161	6 ⁺	0.250%	
		42	41.962402	2 ⁻			51	50.943962	$\frac{7}{2}^{-}$	99.750%	
		43	42.960717	$\frac{3}{2}^{-}$			52	51.944778	3 ⁺	3.76 m (β ⁻)	
		44	43.961560	2 ⁻			53	52.944340	$\frac{7}{2}^{-}$	1.6 m (β ⁻)	
		45	44.960696	$\frac{3}{2}^{+}$			54	53.946442	(3,4,5) ⁺	50 s (β ⁻)	
		46	45.961976	(2 ⁻)		Cr	24	46	45.968360	0 ⁺	0.26 s (ε)
	47	46.961677	$\frac{1}{2}^{+}$	17.5 s (β ⁻)			47	46.962905	$\frac{3}{2}^{-}$	0.51 s (ε)	
Ca	20	38	37.976318	0 ⁺			48	47.954033	0 ⁺	21.6 h (ε)	
		39	38.970718	$\frac{3}{2}^{+}$			49	48.951338	$\frac{5}{2}^{-}$	41.9 m (ε)	
		40	39.962591	0 ⁺			50	49.946046	0 ⁺	4.35%	
		41	40.962278	$\frac{7}{2}^{-}$			51	50.944768	$\frac{7}{2}^{-}$	27.7 d (ε)	
		42	41.958618	0 ⁺			52	51.940510	0 ⁺	83.79%	
		43	42.958766	$\frac{7}{2}^{-}$			53	52.940651	$\frac{3}{2}^{-}$	9.50%	
		44	43.955481	0 ⁺			54	53.938882	0 ⁺	2.36%	
		45	44.956185	$\frac{7}{2}^{-}$			55	54.940842	$\frac{3}{2}^{-}$	3.50 m (β ⁻)	
		46	45.953689	0 ⁺		56	55.940643	$\frac{3}{2}^{-}$	5.9 m (β ⁻)		
		47	46.954543	$\frac{7}{2}^{-}$	Mn	25	50	49.954240	0 ⁺	0.28 s (ε)	
	48	47.952533	0 ⁺	4.54 d (β ⁻)			51	50.948213	$\frac{5}{2}^{-}$	46.2 m (ε)	
Sc	21	42	41.965514	0 ⁺			52	51.945568	6 ⁺	5.59 d (ε)	
		43	42.961150	$\frac{7}{2}^{-}$			53	52.941291	$\frac{7}{2}^{-}$	3.7 My (ε)	
		44	43.959404	2 ⁺			54	53.940361	3 ⁺	312 d (ε)	
		45	44.955910	$\frac{7}{2}^{-}$			55	54.938047	$\frac{5}{2}^{-}$	100%	
		46	45.955170	4 ⁺			56	55.938907	3 ⁺	2.58 h (β ⁻)	
		47	46.952409	$\frac{7}{2}^{-}$			57	56.938285	$\frac{5}{2}^{-}$	1.6 m (β ⁻)	
		48	47.952235	6 ⁺			58	57.940060	3 ⁺	65 s (β ⁻)	
		49	48.950022	$\frac{7}{2}^{-}$		Fe	26	51	50.956825	($\frac{5}{2}$) ⁻	0.25 s (ε)
		50	49.952186	5 ⁺	1.71 m (β ⁻)			52	51.948114	0 ⁺	8.27 h (ε)
	Ti	22	43	42.968523	$\frac{7}{2}^{-}$			53	52.945310	$\frac{7}{2}^{-}$	8.51 m (ε)
		44	43.959690	0 ⁺			54	53.939613	0 ⁺	5.8%	
		45	44.958124	$\frac{7}{2}^{-}$			55	54.938296	$\frac{7}{2}^{-}$	2.7 y (ε)	
		46	45.952629	0 ⁺			56	55.934939	0 ⁺	91.8%	
		47	46.951764	$\frac{5}{2}^{-}$			57	56.935396	$\frac{1}{2}^{-}$	2.15%	
		48	47.947947	0 ⁺			58	57.933277	0 ⁺	0.29%	
		49	48.947871	$\frac{7}{2}^{-}$			59	58.934877	$\frac{5}{2}^{-}$	44.6 d (β ⁻)	
		50	49.944792	0 ⁺			60	59.934078	0 ⁺	1.5 My (β ⁻)	
		51	50.946616	$\frac{7}{2}^{-}$		61	60.936748	($\frac{3}{2}, \frac{5}{2}$) ⁻	6.0 m (β ⁻)		
				5.80 m (β ⁻)		62	61.936773	0 ⁺	68 s (β ⁻)		
					Co	27	54	53.948460	0 ⁺	0.19 s (ε)	

					Abundance or Half-life						
Z	A	Atomic mass (u)	I^{π}	Abundance or Half-life	Z	A	Atomic mass (u)	I^{π}	Abundance or Half-life		
	55	54.942001	$\frac{3}{2}^{-}$	17.5 h (ϵ)	Ga	31	64	63.936836	0^{+}	2.6 m (ϵ)	
	56	55.939841	4^{+}	78.8 d (ϵ)		65	64.932738	$\frac{1}{2}^{-}$	15.2 m (ϵ)		
	57	56.936294	$\frac{3}{2}^{-}$	271 d (ϵ)		66	65.931590	0^{+}	9.4 h (ϵ)		
	58	57.935755	2^{+}	70.8 d (ϵ)		67	66.928204	$\frac{1}{2}^{-}$	78.3 h (ϵ)		
	59	58.933198	$\frac{3}{2}^{-}$	100 %		68	67.927982	1^{+}	68.1 m (ϵ)		
	60	59.933820	5^{+}	5.27 y (β^{-})		69	68.925580	$\frac{3}{2}^{-}$	60.1 %		
	61	60.932478	$\frac{3}{2}^{-}$	1.65 h (β^{-})		70	69.926028	1^{+}	21.1 m (β^{-})		
	62	61.934060	2^{+}	1.5 m (β^{-})		71	70.924701	$\frac{3}{2}^{-}$	39.9 %		
	63	62.933614	$(\frac{7}{2})^{-}$	27.5 s (β^{-})		72	71.926365	3^{-}	14.1 h (β^{-})		
Ni	28	55	54.951336	$\frac{3}{2}^{-}$	0.19 s (ϵ)	73	72.925169	$\frac{3}{2}^{-}$	4.87 h (β^{-})		
		56	55.942134	0^{+}	6.10 d (ϵ)	74	73.926940	$(4)^{-}$	8.1 m (β^{-})		
		57	56.939799	$\frac{3}{2}^{-}$	36.0 h (ϵ)	75	74.926499	$\frac{1}{2}^{-}$	2.1 m (β^{-})		
		58	57.935346	0^{+}	68.3 %	Ge	32	66	65.933847	0^{+}	2.3 h (ϵ)
		59	58.934349	$\frac{3}{2}^{-}$	0.075 My (ϵ)		67	66.932737	$(\frac{1}{2})^{-}$	19.0 m (ϵ)	
		60	59.930788	0^{+}	26.1 %		68	67.928096	0^{+}	271 d (ϵ)	
		61	60.931058	$\frac{3}{2}^{-}$	1.13 %		69	68.927969	$\frac{3}{2}^{-}$	39.0 h (ϵ)	
		62	61.928346	0^{+}	3.59 %		70	69.924250	0^{+}	20.5 %	
		63	62.929670	$\frac{1}{2}^{-}$	100 y (β^{-})		71	70.924954	$\frac{1}{2}^{-}$	11.2 d (ϵ)	
	64	63.927968	0^{+}	0.91 %	72		71.922079	0^{+}	27.4 %		
	65	64.930086	$\frac{3}{2}^{-}$	2.52 h (β^{-})	73		72.923463	$\frac{3}{2}^{+}$	7.8 %		
	66	65.929116	0^{+}	54.8 h (β^{-})	74		73.921177	0^{+}	36.5 %		
	67	66.931570	?	21 s (β^{-})	75	74.922858	$\frac{1}{2}^{-}$	82.8 m (β^{-})			
Cu	29	59	58.939503	$\frac{3}{2}^{-}$	82 s (ϵ)	76	75.921402	0^{+}	7.8 %		
		60	59.937366	2^{+}	23.4 m (ϵ)	77	76.923548	$\frac{3}{2}^{+}$	11.3 h (β^{-})		
		61	60.933461	$\frac{3}{2}^{-}$	3.41 h (ϵ)	78	77.922853	0^{+}	1.45 h (β^{-})		
		62	61.932586	1^{+}	9.73 m (ϵ)	79	78.925360	$(\frac{1}{2})^{-}$	19 s (β^{-})		
		63	62.929599	$\frac{3}{2}^{-}$	69.2 %	As	33	70	69.930929	4^{+}	53 m (ϵ)
		64	63.292766	1^{+}	12.7 h (ϵ)		71	70.927114	$\frac{3}{2}^{-}$	61 h (ϵ)	
		65	64.927793	$\frac{3}{2}^{-}$	30.8 %		72	71.926755	2^{-}	26.0 h (ϵ)	
		66	65.928872	1^{+}	5.10 m (β^{-})		73	72.923827	$\frac{3}{2}^{-}$	80.3 d (ϵ)	
		67	66.927747	$\frac{3}{2}^{-}$	61.9 h (β^{-})		74	73.923928	2^{-}	17.8 d (ϵ)	
	68	67.929620	1^{+}	31 s (β^{-})	75		74.921594	$\frac{1}{2}^{-}$	100 %		
					76		75.922393	2^{-}	26.3 h (β^{-})		
					77		76.920646	$\frac{1}{2}^{-}$	38.8 h (β^{-})		
					78		77.921830	(2^{-})	91 m (β^{-})		
Zn	30	61	60.939514	$\frac{3}{2}^{-}$	89 s (ϵ)	79	78.920946	$\frac{1}{2}^{-}$	9.0 m (β^{-})		
		62	61.934332	0^{+}	9.2 h (ϵ)	Se	34	71	70.932270	$\frac{3}{2}^{-}$	4.7 m (ϵ)
		63	62.933214	$\frac{3}{2}^{-}$	38.1 m (ϵ)		72	71.927110	0^{+}	8.4 d (ϵ)	
		64	63.929145	0^{+}	48.6 %		73	72.926768	$\frac{3}{2}^{+}$	7.1 h (ϵ)	
		65	64.929243	$\frac{3}{2}^{-}$	244 d (ϵ)		74	73.922475	0^{+}	0.87 %	
		66	65.926035	0^{+}	27.9 %		75	74.922522	$\frac{3}{2}^{+}$	119.8 d (ϵ)	
		67	66.927129	$\frac{3}{2}^{-}$	4.10 %		76	75.919212	0^{+}	9.0 %	
		68	67.924846	0^{+}	18.8 %		77	76.919913	$\frac{1}{2}^{-}$	7.6 %	
		69	68.926552	$\frac{1}{2}^{-}$	56 m (β^{-})		78	77.917308	0^{+}	23.5 %	
	70	69.925325	0^{+}	0.62 %	79		78.918498	$\frac{7}{2}^{+}$	< 0.065 My (β^{-})		
	71	70.927727	$\frac{1}{2}^{-}$	2.4 m (β^{-})							
	72	71.926856	0^{+}	46.5 h (β^{-})							
	73	72.929780	$(\frac{3}{2})^{-}$	24 s (β^{-})							

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				Abundance or Half-life				Abundance or Half-life				
Z	A	Atomic mass (u)	I ⁺			Z	A	Atomic mass (u)	I ⁺			
Br 35	76	75.924528	1 ⁻	16.1 h (ε)		87	86.908884	$\frac{1}{2}^{+}$		7.0%		
	77	76.921378	$\frac{1}{2}^{-}$	57.0 h (ε)		88	87.905619	0 ⁺		82.6%		
	78	77.921144	1 ⁺	6.46 m (ε)		89	88.907450	$\frac{1}{2}^{+}$		50.5 d (β ⁻)		
	79	78.918336	$\frac{1}{2}^{-}$	50.69%		90	89.907738	0 ⁺		28.8 y (β ⁻)		
	80	79.918528	1 ⁺	17.6 m (β ⁻)		91	90.910187	$(\frac{2}{1})^{+}$		9.5 h (β ⁻)		
	81	80.916289	$\frac{1}{2}^{-}$	49.31%		92	91.910944	0 ⁺		2.7 h (β ⁻)		
	82	81.916802	$\frac{1}{2}^{-}$	35.3 h (β ⁻)		93	92.913987	$(\frac{1}{2})^{+}$		7.4 m (β ⁻)		
	83	82.915179	$(\frac{2}{1})^{-}$	2.39 h (β ⁻)		Y 39	84	83.920310	(5 ⁻)		39 m (ε)	
	84	83.916503	2 ⁻	31.8 m (β ⁻)			85	84.916437	$(\frac{1}{2})^{-}$		2.7 h (ε)	
	85	84.915612	$(\frac{3}{2})^{-}$	2.9 m (β ⁻)			86	85.914893	4 ⁻		14.7 h (ε)	
Kr 36	75	74.931029	?	4.3 m (ε)			87	86.910882	$\frac{1}{2}^{-}$		80.3 h (ε)	
	76	75.925959	0 ⁺	14.8 h (ε)			88	87.909508	4 ⁻		106.6 d (ε)	
	77	76.924610	$\frac{1}{2}^{+}$	75 m (ε)			89	88.905849	$\frac{1}{2}^{-}$		100%	
	78	77.920396	0 ⁺	0.356%			90	89.907152	2 ⁻		64.1 h (β ⁻)	
	79	78.920084	$\frac{1}{2}^{-}$	35.0 h (ε)			91	90.907303	$\frac{1}{2}^{-}$		58.5 d (β ⁻)	
	80	79.916380	0 ⁺	2.27%			92	91.908917	2 ⁻		3.54 h (β ⁻)	
	81	80.916590	$\frac{1}{2}^{+}$	0.21 My (ε)			93	92.909571	$\frac{1}{2}^{-}$		10.2 h (β ⁻)	
	82	81.913482	0 ⁺	11.6%		94	93.911597	2 ⁻		18.7 m (β ⁻)		
	83	82.914135	$\frac{1}{2}^{+}$	11.5%		Zr 40	87	86.914817	$(\frac{2}{1})^{+}$		1.6 h (ε)	
	84	83.911507	0 ⁺	57.0%			88	87.910225	0 ⁺		83.4 d (ε)	
85	84.912531	$\frac{1}{2}^{+}$	10.7 y (β ⁻)		89		88.908890	$\frac{1}{2}^{+}$		78.4 h (ε)		
86	85.910616	0 ⁺	17.3%		90		89.904703	0 ⁺		51.5%		
87	86.913360	$\frac{1}{2}^{+}$	76 m (β ⁻)		91		90.905644	$\frac{1}{2}^{+}$		11.2%		
88	87.914453	0 ⁺	2.84 h (β ⁻)		92		91.905039	0 ⁺		17.1%		
89	88.917640	$(\frac{1}{2})^{+}$	3.18 m (β ⁻)		93		92.906474	$\frac{1}{2}^{+}$		1.5 My (β ⁻)		
Rb 37	82	81.918195	1 ⁺	1.25 m (ε)			94	93.906315	0 ⁺		17.4%	
	83	82.915144	$\frac{1}{2}^{-}$	86.2 d (ε)			95	94.908042	$\frac{1}{2}^{+}$		64.0 d (β ⁻)	
	84	83.914390	2 ⁻	32.9 d (ε)			96	95.908275	0 ⁺		2.80%	
	85	84.911794	$\frac{1}{2}^{-}$	72.17%		97	96.910950	$\frac{1}{2}^{+}$		16.9 h (β ⁻)		
	86	85.911172	2 ⁻	18.8 d (β ⁻)		98	97.912735	0 ⁺		31 s (β ⁻)		
	87	86.909187	$\frac{1}{2}^{-}$	27.83%		Nb 41	89	88.913449	$(\frac{1}{2})^{-}$		2.0 h (ε)	
	88	87.911326	2 ⁻	17.8 m (β ⁻)			90	89.911263	8 ⁺		14.6 h (ε)	
	89	88.912278	$(\frac{1}{2})^{-}$	15.2 m (β ⁻)			91	90.906991	$(\frac{2}{1})^{+}$		700 y (ε)	
	90	89.914811	(1 ⁻)	153 s (β ⁻)			92	91.907192	(7 ⁺)		35 My (ε)	
	Sr 38	81	80.923270	$(\frac{1}{2})^{-}$	22 m (ε)			93	92.906377	$\frac{1}{2}^{+}$		100%
82		81.918414	0 ⁺	25.0 d (ε)			94	93.907281	6 ⁺		0.020 My (β ⁻)	
83		82.917566	$\frac{1}{2}^{+}$	32.4 d (ε)			95	94.906835	$\frac{1}{2}^{+}$		35.0 d (β ⁻)	
84		83.913430	0 ⁺	0.56%			96	95.908100	6 ⁺		23.4 h (β ⁻)	
85		84.912937	$\frac{1}{2}^{+}$	64.8 d (ε)			97	96.908097	$\frac{1}{2}^{+}$		72 m (β ⁻)	
86		85.909267	0 ⁺	9.8%		Mo 42	90	89.913933	0 ⁺		5.67 h (ε)	
Zr 40		87	86.914817	$(\frac{2}{1})^{+}$			91	90.911755	$\frac{1}{2}^{+}$		15.5 m (ε)	
		88	87.910225	0 ⁺			92	91.906808	0 ⁺		14.8%	
	89	88.908890	$\frac{1}{2}^{+}$		93		92.906813	$\frac{1}{2}^{+}$		3500 y (ε)		
	90	89.904703	0 ⁺		94		93.905085	0 ⁺		9.3%		
	91	90.905644	$\frac{1}{2}^{+}$		95	94.905841	$\frac{1}{2}^{+}$		15.9%			

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Z	A	Atomic mass (u)	I^{π}	Abundance or Half-life	Z	A	Atomic mass (u)	I^{π}	Abundance or Half-life
	96	95.904679	0 ⁺	16.7%		108	107.903895	0 ⁺	26.7%
	97	96.906021	$\frac{1}{2}^{+}$	9.6%		109	108.905954	$\frac{1}{2}^{+}$	13.4 h (β^{-})
	98	97.905407	0 ⁺	24.1%		110	109.905167	0 ⁺	11.8%
	99	98.907711	$\frac{1}{2}^{+}$	66.0 h (β^{-})		111	110.907660	$\frac{1}{2}^{+}$	23 m (β^{-})
	100	99.907477	0 ⁺	9.6%		112	111.907323	0 ⁺	21.0 h (β^{-})
	101	100.910345	$\frac{1}{2}^{+}$	14.6 m (β^{-})	Ag 47	103	102.908980	$\frac{1}{2}^{+}$	65.7 m (ϵ)
Tc 43	94	93.909654	7 ⁺	293 m (ϵ)		104	103.908623	5 ⁺	69.2 m (ϵ)
	95	94.907657	$\frac{3}{2}^{+}$	20.0 h (ϵ)		105	104.906520	$\frac{1}{2}^{+}$	41.3 d (ϵ)
	96	95.907870	7 ⁺	4.3 d (ϵ)		106	105.906662	1 ⁺	24.0 m (ϵ)
	97	96.906364	$\frac{3}{2}^{+}$	2.6 My (ϵ)		107	106.905092	1 ⁺	51.83%
	98	97.907215	(6) ⁺	4.2 My (β^{-})		108	107.905952	1 ⁺	2.4 m (β^{-})
	99	98.906254	$\frac{3}{2}^{+}$	0.214 My (β^{-})		109	108.904756	$\frac{1}{2}^{+}$	48.17%
	100	99.907657	1 ⁺	15.8 s (β^{-})		110	109.906111	1 ⁺	24.4 s (β^{-})
						111	110.905295	$\frac{1}{2}^{+}$	7.45 d (β^{-})
						112	111.907010	2 ⁻	3.14 h (β^{-})
Ru 44	94	93.911361	0 ⁺	52 m (ϵ)					
	95	94.910414	$\frac{1}{2}^{+}$	1.65 h (ϵ)	Cd 48	104	103.909851	0 ⁺	58 m (ϵ)
	96	95.907599	0 ⁺	5.5%		105	104.909459	$\frac{1}{2}^{+}$	56.0 m (ϵ)
	97	96.907556	$\frac{1}{2}^{+}$	2.88 d (ϵ)		106	105.906461	0 ⁺	1.25%
	98	97.905287	0 ⁺	1.86%		107	106.906613	$\frac{1}{2}^{+}$	6.50 h (ϵ)
	99	98.905939	$\frac{1}{2}^{+}$	12.7%		108	107.904176	0 ⁺	0.89%
	100	99.904219	0 ⁺	12.6%		109	108.904953	$\frac{1}{2}^{+}$	463 d (ϵ)
	101	100.905582	$\frac{1}{2}^{+}$	17.0%		110	109.903005	0 ⁺	12.5%
	102	101.904348	0 ⁺	31.6%		111	110.904182	1 ⁺	12.8%
	103	102.906323	$\frac{1}{2}^{+}$	39.4 d (β^{-})		112	111.902757	0 ⁺	24.1%
	104	103.905424	0 ⁺	18.7%		113	112.904400	$\frac{1}{2}^{+}$	12.2%
	105	104.907744	$\frac{1}{2}^{+}$	4.44 h (β^{-})		114	113.903357	0 ⁺	28.7%
	106	105.907321	0 ⁺	372 d (β^{-})		115	114.905430	$\frac{1}{2}^{+}$	53.4 h (β^{-})
	107	106.910130	($\frac{1}{2}^{+}$)	3.8 m (β^{-})		116	115.904755	0 ⁺	7.5%
						117	116.907228	$\frac{1}{2}^{+}$	2.4 h (β^{-})
						118	117.911700	0 ⁺	50.3 m (β^{-})
Rh 45	98	97.910716	(2) ⁺	8.7 m (ϵ)					
	99	98.908192	($\frac{1}{2}^{+}$)	16.1 d (ϵ)	In 49	110	109.907230	2 ⁺	69.1 m (ϵ)
	100	99.908116	1 ⁻	20.8 h (ϵ)		111	110.905109	$\frac{1}{2}^{+}$	2.83 d (ϵ)
	101	100.906159	$\frac{1}{2}^{+}$	3.3 y (ϵ)		112	111.905536	1 ⁺	14.4 m (ϵ)
	102	101.906814	6 ⁺	2.9 y (ϵ)		113	112.904061	$\frac{1}{2}^{+}$	4.3%
	103	102.905500	$\frac{1}{2}^{+}$	100%		114	113.904916	1 ⁺	71.9 s (β^{-})
	104	103.906651	1 ⁺	42.3 s (β^{-})		115	114.903882	$\frac{1}{2}^{+}$	95.7%
	105	104.905686	$\frac{1}{2}^{+}$	35.4 h (β^{-})		116	115.905264	1 ⁺	14.1 s (β^{-})
	106	105.907279	1 ⁺	29.8 s (β^{-})		117	116.904517	$\frac{1}{2}^{+}$	43.8 m (β^{-})
Pd 46	99	98.911763	($\frac{1}{2}^{+}$)	21.4 m (ϵ)					
	100	99.908527	0 ⁺	3.6 d (ϵ)	Sn 50	109	108.911294	$\frac{1}{2}^{+}$	18.0 m (ϵ)
	101	100.908287	$\frac{1}{2}^{+}$	8.5 h (ϵ)		110	109.907858	0 ⁺	4.1 h (ϵ)
	102	101.905634	0 ⁺	1.0%		111	110.907741	$\frac{1}{2}^{+}$	35 m (ϵ)
	103	102.906114	$\frac{1}{2}^{+}$	17.0 d (ϵ)		112	111.904826	0 ⁺	1.01%
	104	103.904029	0 ⁺	11.0%		113	112.905176	$\frac{1}{2}^{+}$	115.1 d (ϵ)
	105	104.905079	$\frac{1}{2}^{+}$	22.2%		114	113.902784	0 ⁺	0.67%
	106	105.903478	0 ⁺	27.3%		115	114.903348	$\frac{1}{2}^{+}$	0.38%
	107	106.905127	$\frac{1}{2}^{+}$	6.5 My (β^{-})					

				Abundance or Half-life					Abundance or Half-life
Z	A	Atomic mass (u)	I^π		Z	A	Atomic mass (u)	I^π	
		116 115.901747	0 ⁺	14.6%		129	128.904986	$\frac{7}{2}^+$	16 My (β^-)
		117 116.902956	$\frac{1}{2}^+$	7.75%		130	129.906713	5 ⁺	12.4 h (β^-)
		118 117.901609	0 ⁺	24.3%		131	130.906114	$\frac{3}{2}^+$	8.04 d (β^-)
		119 118.903311	$\frac{1}{2}^+$	8.6%		132	131.907987	4 ⁺	2.30 h (β^-)
		120 119.902199	0 ⁺	32.4%					
		121 120.904239	$\frac{3}{2}^+$	27.1 h (β^-)	Xe	54	121 120.911450	($\frac{5}{2}^+$)	40.1 m (ϵ)
		122 121.903440	0 ⁺	4.56%		122	121.908170	0 ⁺	20.1 h (ϵ)
		123 122.905722	$\frac{11}{2}^-$	129 d (β^-)		123	122.908469	($\frac{5}{2}^+$)	2.08 h (ϵ)
		124 123.905274	0 ⁺	5.64%		124	123.905894	0 ⁺	0.096%
		125 124.907785	$\frac{11}{2}^-$	9.62 d (β^-)		125	124.906397	($\frac{5}{2}^+$)	17 h (ϵ)
		126 125.907654	0 ⁺	0.1 My (β^-)		126	125.904281	0 ⁺	0.090%
		127 126.910355	($\frac{11}{2}^-$)	2.1 h (β^-)		127	126.905182	($\frac{5}{2}^+$)	36.4 d (ϵ)
						128	127.903531	0 ⁺	1.92%
Sb	51	118 117.905534	1 ⁺	3.6 m (ϵ)		129	128.904780	$\frac{1}{2}^+$	26.4%
		119 118.903948	$\frac{3}{2}^+$	38.0 h (ϵ)		130	129.903509	0 ⁺	4.1%
		120 119.905077	1 ⁺	15.8 m (ϵ)		131	130.905072	$\frac{3}{2}^+$	21.2%
		121 120.903821	$\frac{5}{2}^+$	57.3%		132	131.904144	0 ⁺	26.9%
		122 121.905179	2 ⁻	2.70 d (β^-)		133	132.905888	$\frac{3}{2}^+$	5.25 d (β^-)
		123 122.904216	$\frac{3}{2}^+$	42.7%		134	133.905395	0 ⁺	10.4%
		124 123.905938	3 ⁻	60.2 d (β^-)		135	134.907130	$\frac{3}{2}^+$	9.1 h (β^-)
		125 124.905252	$\frac{3}{2}^+$	2.7 y (β^-)		136	135.907214	0 ⁺	8.9%
		126 125.907250	8 ⁻	12.4 d (β^-)		137	136.911557	$\frac{7}{2}^-$	3.82 m (β^-)
		127 126.906919	$\frac{7}{2}^+$	3.85 d (β^-)					
					Cs	55	130 129.906753	1 ⁺	29.2 m (ϵ)
Te	52	117 116.908630	$\frac{1}{2}^+$	62 m (ϵ)		131	130.905444	$\frac{3}{2}^+$	9.69 d (ϵ)
		118 117.905908	0 ⁺	6.00 d (ϵ)		132	131.906431	2 ⁻	6.47 d (ϵ)
		119 118.906411	$\frac{1}{2}^+$	16.0 h (ϵ)		133	132.905429	$\frac{3}{2}^+$	100%
		120 119.904048	0 ⁺	0.091%		134	133.906696	4 ⁺	2.06 y (β^-)
		121 120.904947	$\frac{1}{2}^+$	16.8 d (ϵ)		135	134.905885	$\frac{3}{2}^+$	3 My (β^-)
		122 121.903050	0 ⁺	2.5%		136	135.907289	5 ⁺	13.1 d (β^-)
		123 122.904271	$\frac{1}{2}^+$	0.89%		137	136.907073	$\frac{7}{2}^+$	30.2 y (β^-)
		124 123.902818	0 ⁺	4.6%		138	137.911004	3 ⁻	32.2 m (β^-)
		125 124.904429	$\frac{1}{2}^+$	7.0%					
		126 125.903310	0 ⁺	18.7%	Ba	56	127 126.911130	($\frac{5}{2}^+$)	12.7 m (ϵ)
		127 126.905221	$\frac{3}{2}^+$	9.4 h (β^-)		128	127.908237	0 ⁺	2.43 d (ϵ)
		128 127.904463	0 ⁺	31.7%		129	128.908642	$\frac{1}{2}^+$	2.2 h (ϵ)
		129 128.906594	$\frac{3}{2}^+$	69 m (β^-)		130	129.906282	0 ⁺	0.106%
		130 129.906229	0 ⁺	34.5%		131	130.906902	$\frac{1}{2}^+$	12.0 d (ϵ)
		131 130.908528	$\frac{1}{2}^+$	25.0 m (β^-)		132	131.905042	0 ⁺	0.101%
		132 131.908517	0 ⁺	78.2 h (β^-)		133	132.905988	$\frac{1}{2}^+$	10.7 y (ϵ)
		133 132.910910	($\frac{5}{2}^+$)	12.5 m (β^-)		134	133.904486	0 ⁺	2.42%
						135	134.905665	$\frac{1}{2}^+$	6.59%
I	53	123 122.905594	$\frac{3}{2}^+$	13.2 h (ϵ)		136	135.904553	0 ⁺	7.85%
		124 123.906207	2 ⁻	4.18 d (ϵ)		137	136.905812	$\frac{1}{2}^+$	11.2%
		125 124.904620	$\frac{3}{2}^+$	60.2 d (ϵ)		138	137.905232	0 ⁺	71.7%
		126 125.905624	2 ⁻	13.0 d (ϵ)		139	138.908826	$\frac{7}{2}^-$	82.9 m (β^-)
		127 126.904473	$\frac{3}{2}^+$	100%		140	139.910581	0 ⁺	12.7 d (β^-)
		128 127.905810	1 ⁺	25.0 m (β^-)		141	140.914363	$\frac{3}{2}^-$	18.3 m (β^-)

TABLE OF NUCLEAR PROPERTIES 1829

				Abundance or Half-life					Abundance or Half-life
Z	A	Atomic mass (u)	I^{π}		Z	A	Atomic mass (u)	I^{π}	
La 57	135	134.906953	$\frac{5}{2}^{+}$	19.5 h (ϵ)		145	144.912743	$\frac{5}{2}^{+}$	17.7 y (ϵ)
	136	135.907630	1^{+}	9.87 m (ϵ)		146	145.914708	3^{-}	5.5 y (ϵ)
	137	136.906460	$\frac{7}{2}^{+}$	0.06 My (ϵ)		147	146.915135	$\frac{7}{2}^{+}$	2.62 y (β^{-})
	138	137.907105	5^{+}	0.089%		148	147.917473	1^{-}	5.37 d (β^{-})
	139	138.906347	$\frac{7}{2}^{+}$	99.911%		149	148.918332	$\frac{7}{2}^{+}$	53.1 h (β^{-})
	140	139.909471	3^{-}	40.3 h (β^{-})		150	149.920981	(1^{-})	2.68 h (β^{-})
	141	140.910896	$\frac{7}{2}^{+}$	3.90 h (β^{-})	Sm 62	142	141.915206	0^{+}	72.5 m (ϵ)
	142	141.914090	2^{-}	91.1 m (β^{-})		143	142.914626	$\frac{5}{2}^{+}$	8.83 m (ϵ)
Ce 58	133	132.911360	$\frac{1}{2}^{+}$	5.4 h (ϵ)		144	143.911998	0^{+}	3.1%
	134	133.908890	0^{+}	76 h (ϵ)		145	144.913409	$\frac{1}{2}^{-}$	340 d (ϵ)
	135	134.909117	$\frac{1}{2}^{+}$	17.6 h (ϵ)		146	145.913053	0^{+}	103 My (α)
	136	135.907140	0^{+}	0.190%		147	146.914894	$\frac{1}{2}^{-}$	15.1%
	137	136.907780	$\frac{3}{2}^{+}$	9.0 h (ϵ)		148	147.914819	0^{+}	11.3%
	138	137.905985	0^{+}	0.254%		149	148.917180	$\frac{3}{2}^{-}$	13.9%
	139	138.906631	$\frac{1}{2}^{+}$	137.2 d (ϵ)		150	149.917273	0^{+}	7.4%
	140	139.905433	0^{+}	88.5%		151	150.919929	$\frac{3}{2}^{-}$	90 y (β^{-})
	141	140.908271	$\frac{1}{2}^{-}$	32.5 d (β^{-})		152	151.919728	0^{+}	26.6%
	142	141.909241	0^{+}	11.1%		153	152.922094	$\frac{3}{2}^{+}$	46.8 h (β^{-})
	143	142.912383	$\frac{1}{2}^{-}$	33.0 h (β^{-})		154	153.922205	0^{+}	22.6%
	144	143.913643	0^{+}	284 d (β^{-})		155	154.924636	$\frac{1}{2}^{-}$	22.4 m (β^{-})
	145	144.917230	$\frac{3}{2}^{+}$	2.98 m (β^{-})	Eu 63	148	147.918125	5^{-}	54.5 d (ϵ)
Pr 59	138	137.910748	1^{+}	1.45 m (ϵ)		149	148.917926	$\frac{5}{2}^{+}$	93.1 d (ϵ)
	139	138.908917	$\frac{3}{2}^{+}$	4.4 h (ϵ)		150	149.919702	0^{-}	36 y (ϵ)
	140	139.909071	1^{+}	3.39 m (ϵ)		151	150.919847	$\frac{5}{2}^{+}$	47.9%
	141	140.907647	$\frac{3}{2}^{+}$	100%		152	151.921742	3^{-}	13 y (ϵ)
	142	141.910039	2^{-}	19.2 h (β^{-})		153	152.921225	$\frac{5}{2}^{+}$	52.1%
	143	142.910814	$\frac{1}{2}^{+}$	13.6 d (β^{-})		154	153.922975	3^{-}	8.5 y (β^{-})
	144	143.913301	0^{-}	17.3 m (β^{-})		155	154.922889	$\frac{5}{2}^{+}$	4.9 y (β^{-})
						156	155.924752	0^{+}	15 d (β^{-})
Nd 60	139	138.911920	$\frac{1}{2}^{+}$	29.7 m (ϵ)		157	156.925418	$\frac{5}{2}^{+}$	15 h (β^{-})
	140	139.909306	0^{+}	3.37 d (ϵ)	Gd 64	149	148.919344	$\frac{7}{2}^{-}$	9.4 d (ϵ)
	141	140.909594	$\frac{1}{2}^{+}$	2.5 h (ϵ)		150	149.918662	0^{+}	1.8 My (α)
	142	141.907719	0^{+}	27.2%		151	150.920346	$\frac{7}{2}^{-}$	120 d (ϵ)
	143	142.909810	$\frac{1}{2}^{-}$	12.2%		152	151.919786	0^{+}	0.20%
	144	143.910083	0^{+}	23.8%		153	152.921745	$\frac{7}{2}^{-}$	242 d (ϵ)
	145	144.912570	$\frac{1}{2}^{-}$	8.3%		154	153.920861	0^{+}	2.1%
	146	145.913113	0^{+}	17.2%		155	154.922618	$\frac{7}{2}^{-}$	14.8%
	147	146.916097	$\frac{3}{2}^{-}$	11.0 d (β^{-})		156	155.922118	0^{+}	20.6%
	148	147.916889	0^{+}	5.7%		157	156.923956	$\frac{7}{2}^{-}$	15.7%
	149	148.920145	$\frac{5}{2}^{-}$	1.73 h (β^{-})		158	157.924099	0^{+}	24.8%
	150	149.920887	0^{+}	5.6%		159	158.926384	$\frac{7}{2}^{-}$	18.6 h (β^{-})
	151	150.923825	$(\frac{1}{2}^{+})$	12.4 m (β^{-})		160	159.927049	0^{+}	21.8%
	152	151.924680	0^{+}	11.4 m (β^{-})		161	160.929664	$\frac{7}{2}^{-}$	3.7 m (β^{-})
Pm 61	142	141.912970	1^{+}	40.5 s (ϵ)	Tb 65	156	155.924742	3^{-}	5.34 d (ϵ)
	143	142.910930	$\frac{1}{2}^{+}$	265 d (ϵ)		157	156.924023	$\frac{5}{2}^{+}$	150 y (ϵ)
	144	143.912588	5^{-}	349 d (ϵ)					

Z		Atomic mass (u)	Abundance or Half-life	Z		Atomic mass (u)	Abundance or Half-life				
Yb	70	166	165.933875	0 ⁺	56.7 h (e)	Lu	71	172	171.939085	(4 ⁻)	6.70 d (e)
		167	166.934946	$\frac{1}{2}^{-}$	17.5 m (e)			173	172.938929	$\frac{1}{2}^{-}$	1.37 y (e)
		168	167.933894	0 ⁺	0.135%			174	173.940336	1 ⁻	3.3 y (e)
		169	168.935186	$\frac{1}{2}^{-}$	32.0 d (e)			175	174.940770	$\frac{1}{2}^{-}$	97.39%
		170	169.934759	0 ⁺	3.1%			176	175.942679	7 ⁻	2.61%
		171	170.936323	$\frac{1}{2}^{-}$	14.4%			177	176.943752	$\frac{1}{2}^{-}$	6.71 d (e)
		172	171.936378	0 ⁺	21.9%			178	177.945963	1 ⁺	28.4 m (e)
		173	172.938208	$\frac{1}{2}^{-}$	16.2%						
		174	173.938859	0 ⁺	31.6%						
		175	174.941273	$\frac{1}{2}^{-}$	4.19 d (e)						
Dy	66	153	152.925769	$\frac{1}{2}^{-}$	6.4 h (e)	Hf	72	171	170.940490	($\frac{1}{2}^{-}$)	12.1 h (e)
		154	153.924429	0 ⁺	3 My (a)			172	171.939460	0 ⁺	1.87 y (e)
		155	154.925747	$\frac{1}{2}^{-}$	10.0 h (e)			173	172.940650	$\frac{1}{2}^{-}$	24.0 h (e)
		156	155.924277	0 ⁺	0.057%			174	173.940044	0 ⁺	0.16%
		157	156.925460	$\frac{1}{2}^{-}$	8.1 h (e)			175	174.941507	$\frac{1}{2}^{-}$	70 d (e)
		158	157.924403	0 ⁺	0.100%			176	175.941406	0 ⁺	5.2%
		159	158.925735	$\frac{1}{2}^{-}$	144.4 d (e)			177	176.943217	$\frac{1}{2}^{-}$	18.6%
		160	159.925193	0 ⁺	2.3%			178	177.943696	0 ⁺	27.1%
		161	160.926930	$\frac{1}{2}^{-}$	19.90%			179	178.945812	$\frac{1}{2}^{-}$	13.7%
		162	161.926795	0 ⁺	25.5%			180	179.946546	0 ⁺	35.2%
Ho	67	162	161.928728	$\frac{1}{2}^{-}$	24.9%	Ta	73	178	177.945750	1 ⁺	9.31 m (e)
		163	162.928728	$\frac{1}{2}^{-}$	28.1%			179	178.945930	($\frac{1}{2}^{-}$)	665 d (e)
		164	163.929171	0 ⁺	2.33 h (e)			180	179.947462	1 ⁺	0.0123%
		165	164.931700	$\frac{1}{2}^{-}$	81.6 h (e)			181	180.947992	$\frac{1}{2}^{-}$	99.9877%
		166	165.932803	0 ⁺	26.8 h (e)			182	181.950149	3 ⁻	115 d (e)
		167	166.933127	($\frac{1}{2}^{-}$)	3.1 h (e)			183	182.951369	$\frac{1}{2}^{-}$	5.1 d (e)
		168	167.933268	0 ⁺	27.1%	W	74	178	177.945840	0 ⁺	21.5 d (e)
		169	168.934588	$\frac{1}{2}^{-}$	9.40 d (e)			179	178.947067	($\frac{1}{2}^{-}$)	38 m (e)
		170	169.935461	0 ⁺	14.9%			180	179.946701	0 ⁺	0.13%
		171	170.938027	$\frac{1}{2}^{-}$	7.52 h (e)			181	180.948192	$\frac{1}{2}^{-}$	121 d (e)
		172	171.939353	0 ⁺	49.3 h (e)			182	181.948202	0 ⁺	26.3%
		173	172.940679	$\frac{1}{2}^{-}$	22.9%			183	182.950220	$\frac{1}{2}^{-}$	14.3%
		174	173.941999	0 ⁺	27.1%						
		175	174.943319	$\frac{1}{2}^{-}$	10.4 h (e)						
		176	175.944639	0 ⁺	33.4%						
		177	176.945959	$\frac{1}{2}^{-}$	22.9%						
Er	68	160	159.929080	0 ⁺	28.6 h (e)	Tm	69	166	165.933561	2 ⁺	7.70 h (e)
		161	160.929996	$\frac{1}{2}^{-}$	3.24 h (e)			167	166.932848	1 ⁺	9.25 d (e)
		162	161.928775	0 ⁺	0.14%			168	167.934170	3 ⁺	93.1 d (e)
		163	162.930030	$\frac{1}{2}^{-}$	75.1 m (e)			169	168.934212	1 ⁺	100%
		164	163.929198	0 ⁺	1.56%			170	169.935798	1 ⁻	128.6 d (e)
		165	164.930723	$\frac{1}{2}^{-}$	10.4 h (e)			171	170.936427	$\frac{1}{2}^{-}$	1.92 y (e)
		166	165.930290	0 ⁺	33.4%			172	171.938397	2 ⁺	63.6 h (e)
		167	166.932046	$\frac{1}{2}^{-}$	22.9%						
		168	167.932368	0 ⁺	27.1%						
		169	168.934588	$\frac{1}{2}^{-}$	9.40 d (e)						

Z	A	Atomic mass (u)	I^π	Abundance or Half-life	Z	A	Atomic mass (u)	I^π	Abundance or Half-life
	184	183.950928	0 ⁺	30.7%		198	197.967869	0 ⁺	7.2%
	185	184.953416	$\frac{1}{2}^-$	75.1 d (β^-)		199	198.970552	($\frac{1}{2}^-$)	30.8 m (β^-)
	186	185.954357	0 ⁺	28.6%		200	199.971417	0 ⁺	12.5 h (β^-)
	187	186.957153	$\frac{1}{2}^-$	23.9 h (β^-)					
	188	187.958480	0 ⁺	69.4 d (β^-)	Au 79	194	193.965348	1 ⁻	39.5 h (e)
Re 75	182	181.951210	2 ⁺	12.7 h (e)		195	194.965013	$\frac{1}{2}^+$	186 d (e)
	183	182.950817	($\frac{1}{2}$) ⁺	71 d (e)		196	195.966544	2 ⁻	6.18 d (e)
	184	183.952530	3 ⁻	38 d (e)		197	196.966543	$\frac{1}{2}^+$	100%
	185	184.952951	$\frac{3}{2}^+$	37.40%		198	197.968217	2 ⁻	2.696 d (β^-)
	186	185.954984	1 ⁻	90.6 h (β^-)		199	198.968740	$\frac{1}{2}^+$	3.14 d (β^-)
	187	186.955744	$\frac{1}{2}^+$	62.60%		200	199.970670	1 ⁻	48.4 m (β^-)
	188	187.958106	1 ⁻	16.9 h (β^-)					
	189	188.959219	($\frac{1}{2}$) ⁺	24.3 h (β^-)	Hg 80	193	192.966560	$\frac{1}{2}^-$	3.8 h (e)
Os 76	182	181.952120	0 ⁺	21.5 h (e)		194	193.965391	0 ⁺	520 y (e)
	183	182.953290	($\frac{1}{2}$) ⁺	13.0 h (e)		195	194.966640	$\frac{1}{2}^-$	9.5 h (e)
	184	183.952488	0 ⁺	0.018%		196	195.965807	0 ⁺	0.15%
	185	184.954041	$\frac{1}{2}^-$	93.6 d (e)		197	196.967187	$\frac{1}{2}^-$	64.1 h (e)
	186	185.953830	0 ⁺	1.6%		198	197.966743	0 ⁺	10.0%
	187	186.955741	$\frac{1}{2}^-$	1.6%		199	198.968254	$\frac{1}{2}^-$	16.8%
	188	187.955830	0 ⁺	13.3%		200	199.968300	0 ⁺	23.1%
	189	188.958137	$\frac{1}{2}^-$	16.1%		201	200.970277	$\frac{1}{2}^-$	13.2%
	190	189.958436	0 ⁺	26.4%		202	201.970617	0 ⁺	29.8%
	191	190.960920	$\frac{3}{2}^-$	15.4 d (β^-)		203	202.972848	$\frac{1}{2}^-$	46.6 d (β^-)
	192	191.961467	0 ⁺	41.0%		204	203.973467	0 ⁺	6.9%
	193	192.964138	$\frac{1}{2}^-$	30.6 h (β^-)		205	204.976047	$\frac{1}{2}^-$	5.2 m (β^-)
	194	193.965173	0 ⁺	6.0 y (β^-)	Tl 81	200	199.970934	2 ⁻	26.1 h (e)
Ir 77	188	187.958830	(2 ⁻)	41.5 h (e)		201	200.970794	$\frac{1}{2}^+$	73 h (e)
	189	188.958712	$\frac{1}{2}^+$	13.1 d (e)		202	201.972085	2 ⁻	12.2 d (e)
	190	189.960580	(4 ⁺)	11.8 d (e)		203	202.972320	$\frac{1}{2}^+$	29.5%
	191	190.960584	$\frac{1}{2}^+$	37.3%		204	203.973839	2 ⁻	3.77 y (β^-)
	192	191.962580	4 ⁻	74.2 d (β^-)		205	204.974401	$\frac{1}{2}^+$	70.5%
	193	192.962917	$\frac{1}{2}^+$	62.7%		206	205.976084	0 ⁻	4.20 m (β^-)
	194	193.965069	1 ⁻	19.2 h (β^-)	Pb 82	201	200.972830	$\frac{1}{2}^-$	9.3 h (e)
	195	194.965966	($\frac{1}{2}$) ⁺	2.8 h (β^-)		202	201.972134	0 ⁺	0.05 My (e)
Pt 78	187	186.960470	$\frac{1}{2}^-$	2.35 h (e)		203	202.973365	$\frac{1}{2}^-$	51.9 h (e)
	188	187.959386	0 ⁺	10.2 d (e)		204	203.973020	0 ⁺	1.42%
	189	188.960817	$\frac{1}{2}^-$	10.9 h (e)		205	204.974458	$\frac{1}{2}^-$	15 My (e)
	190	189.959917	0 ⁺	0.013%		206	205.974440	0 ⁺	24.1%
	191	190.961665	$\frac{1}{2}^-$	2.9 d (e)		207	206.975872	$\frac{1}{2}^-$	22.1%
	192	191.961019	0 ⁺	0.78%		208	207.976627	0 ⁺	52.3%
	193	192.962977	($\frac{1}{2}$) ⁻	50 y (e)		209	208.981065	$\frac{1}{2}^+$	3.25 h (β^-)
	194	193.962655	0 ⁺	32.9%		210	209.984163	0 ⁺	22.3 y (β^-)
	195	194.964766	$\frac{1}{2}^-$	33.8%		211	210.988735	($\frac{1}{2}$) ⁺	36.1 m (β^-)
	196	195.964926	0 ⁺	25.3%		212	211.991871	0 ⁺	10.6 h (β^-)
	197	196.967315	$\frac{1}{2}^-$	18.3 h (β^-)	Bi 83	206	205.978478	6 ⁺	6.24 d (e)
						207	206.978446	$\frac{1}{2}^-$	32 y (e)

Abundance or Half-life					Abundance or Half-life				
Z	A	Atomic mass (u)	I ^π	Half-life	Z	A	Atomic mass (u)	I ^π	Half-life
	208	207.979717	(5 ⁺)	0.368 My (ε)		232	232.038051	0 ⁺	100 %
	209	208.980374	$\frac{9}{2}^{-}$	100 %		233	233.041577	($\frac{1}{2}^{+}$)	22.3 m (β ⁻)
	210	209.984095	1 ⁻	5.01 d (β ⁻)					
	211	210.987255	$\frac{9}{2}^{-}$	2.15 m (α)	Pa 91	229	229.032073	($\frac{5}{2}^{+}$)	1.4 d (ε)
	212	211.991255	1 ⁻	60.6 m (β ⁻)		230	230.034527	(2 ⁻)	17.7 d (ε)
Po 84	206	205.980456	0 ⁺	8.8 d (ε)		231	231.035880	$\frac{3}{2}^{-}$	32,800 y (α)
	207	206.981570	$\frac{5}{2}^{-}$	5.8 h (ε)		232	232.038565	(2 ⁻)	1.31 d (β ⁻)
	208	207.981222	0 ⁺	2.90 y (α)		233	233.040243	$\frac{3}{2}^{-}$	27.0 d (β ⁻)
	209	208.982404	$\frac{1}{2}^{-}$	102 y (α)					
	210	209.982848	0 ⁺	138.4 d (α)	U 92	233	233.039628	$\frac{5}{2}^{+}$	0.1592 My (α)
	211	210.986627	$\frac{9}{2}^{+}$	0.52 s (α)		234	234.040947	0 ⁺	0.245 My (α)
						235	235.043924	$\frac{7}{2}^{-}$	0.720 %
At 85	208	207.986510	6 ⁺	1.63 h (ε)		236	236.045563	0 ⁺	23.42 My (α)
	209	208.986149	$\frac{9}{2}^{-}$	5.4 h (ε)		237	237.048725	$\frac{1}{2}^{+}$	6.75 d (β ⁻)
	210	209.987126	5 ⁺	8.3 h (ε)		238	238.050785	0 ⁺	99.275 %
	211	210.987469	$\frac{9}{2}^{-}$	7.21 h (ε)		239	239.054290	$\frac{5}{2}^{+}$	23.5 m (β ⁻)
	212	211.990725	(1 ⁻)	0.31 s (α)					
	213	212.992911	$\frac{9}{2}^{-}$	0.11 μs (α)	Np 93	236	236.046550	(6 ⁻)	0.11 My (ε)
						237	237.048168	$\frac{5}{2}^{+}$	2.14 My (α)
Rn 86	207	206.990690	$\frac{5}{2}^{-}$	9.3 m (ε)		238	238.050941	2 ⁺	2.117 d (β ⁻)
	210	209.989669	0 ⁺	2.4 h (α)		239	239.052933	$\frac{5}{2}^{+}$	2.36 d (β ⁻)
	211	210.990576	$\frac{1}{2}^{-}$	14.6 h (ε)					
	212	211.990697	0 ⁺	24 m (α)	Pu 94	237	237.048401	$\frac{7}{2}^{-}$	45.3 d (ε)
	218	218.005580	0 ⁺	35 ms (α)		238	238.049555	0 ⁺	87.74 y (α)
	222	222.017571	0 ⁺	3.82 d (α)		239	239.052158	$\frac{1}{2}^{+}$	24,100 y (α)
	224		0 ⁺	107 m (β ⁻)		240	240.053808	0 ⁺	6570 y (α)
						241	241.056846	$\frac{5}{2}^{+}$	14.4 y (β ⁻)
Fr 87	209	208.995870	$\frac{9}{2}^{-}$	50 s (α)		242	242.058737	0 ⁺	0.376 My (α)
	212	211.996130	5 ⁺	20 m (ε)		243	243.061998	$\frac{7}{2}^{+}$	4.96 h (β ⁻)
	215	215.000310	$\frac{9}{2}^{-}$	0.12 μs (α)					
	220	220.012293	1	27.4 s (α)	Am 95	240	240.055278	(3 ⁻)	50.9 h (ε)
	223	223.019733	($\frac{3}{2}^{-}$)	21.8 m (β ⁻)		241	241.056824	$\frac{5}{2}^{+}$	433 y (α)
						242	242.059542	1 ⁻	16.0 h (β ⁻)
Ra 88	222	222.015353	0 ⁺	38 s (α)		243	243.061375	$\frac{5}{2}^{-}$	7370 y (α)
	223	223.018501	$\frac{1}{2}^{+}$	11.4 d (α)		244	244.064279	(6 ⁻)	10.1 h (β ⁻)
	224	224.020186	0 ⁺	3.66 d (α)					
	225	225.023604	($\frac{3}{2}^{+}$)	14.8 d (β ⁻)	Cm 96	246	246.067218	0 ⁺	4700 y (α)
	226	226.025403	0 ⁺	1602 y (α)		247	247.070347	$\frac{3}{2}^{-}$	16 My (α)
	227	227.029171	($\frac{3}{2}^{+}$)	42 m (β ⁻)		248	248.072343	0 ⁺	0.34 My (α)
						249	249.075948	$\frac{1}{2}^{+}$	64 m (β ⁻)
Ac 89	224	224.021685	(0 ⁻)	2.9 h (ε)					
	225	225.023205	($\frac{3}{2}^{-}$)	10.0 d (α)	Bk 97	246	246.068720	2 ⁻	1.8 d (ε)
	226	226.026084	(1 ⁻)	29 h (β ⁻)		247	247.070300	($\frac{3}{2}^{-}$)	1380 y (α)
	227	227.027750	$\frac{1}{2}^{-}$	21.77 y (β ⁻)					
	228	228.031015	(3 ⁺)	6.1 h (β ⁻)	Cf 98	251	251.079580	$\frac{1}{2}^{+}$	898 y (α)
						252	252.081621	0 ⁺	2.64 y (α)
Th 90	228	228.028715	0 ⁺	1.91 y (α)					
	229	229.031755	$\frac{5}{2}^{+}$	7300 y (α)	Es 99	252	252.082944	(4 ⁺ , 5 ⁻)	472 d (α)
	230	230.033128	0 ⁺	75,400 y (α)		253	253.084818	$\frac{7}{2}^{+}$	20.5 d (α)
	231	231.036299	$\frac{5}{2}^{+}$	25.52 h (β ⁻)					