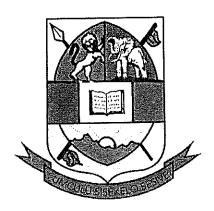
UNIVERSITY OF ESWATINI



Final Examination

TITLE OF PAPER:

Natural Products Chemistry

COURSE NUMBER:

CHE 432

TIME ALLOWED:

Three Hours

INSTRUCTIONS:

Answer any four (4) questions of the six (6) questions and every question holds 25 marks. NB: all questions are to be answered in a separate answer sheet.

Question 1

- a) Explain clearly the primary, secondary, tertiary and quaternary structures of proteins.(12)
- b) What are enzymes and what is their role in the body of higher animals? (6)
- c) Give examples of acidic and basic amino acids including their structure? (7)

Question 2

- a) What is the meaning of reducing sugars? What are Tollen's reagent and Fehling's solution and what are they used for? (10)
- b) What are carbohydrates? Give the functional and structural similarity and difference between the three carbohydrate polymers starch, glycogen and cellulose. (10)
- c) Explain Ruff degradation reaction?(5)

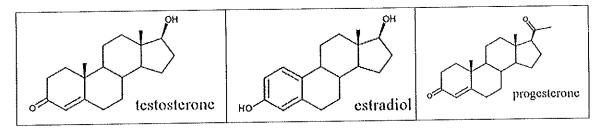
Question 3

- a) The following reactions are used as qualitative tests of Alkaloids. Explain the reactions including their mechanism. (6)
 - i) Dragendorff's reagent
 - ii) Mayer's reagent
- b) Show the biosynthetic route for the production of alkaloids, terpenoids and flavonoids taking into account the most basic chemicals (H₂O and CO₂). (10)
- c) Is the compound given here an alkaloid? Explain and justify your answer. (5)

d) Mention the two commonly known malaria drugs which are alkaloids? (4)

Question 4

a) What are the active functional groups in the following hormones? (10)

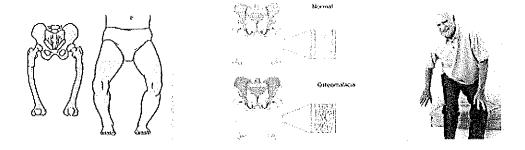


- b) What are the roles of these hormones in determining sexual characteristics? (9)
- c) What are the types of hormones in human bodies and how they transported? (6)

Question 5

a) How is cholesterol transformed to calciferol? (6)

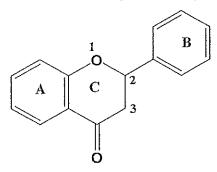
b) What can you observe from the following picture? (6)



c) Show that cholesterol is a Triterpen using the isoprene rule. (5)

Question (6)

The flavone indicated below has three important component parts rings A, B and C.



- a) Which of the three rings is more reactive? Give reasons for your choice? (6)
- b) How is the chemical properties modified when 2-3 double bonds are introduced? (6)
- c) Mention some common sources of flavonoids. (5)
- d) Why are flavanoids very useful in medicinal use? Explain and justify your answer. (8)