

UNIVERSITY OF SWAZILAND Faculty of Health Sciences Department of Environmental Health Science

DEGREE IN: - ENVIRONMENTAL MANAGEMENT & OCCUPATIONAL SAFETY AND HEALTH - ENVIRONMENTAL

MANAGEMENT AND WATER RESOURCES

SUPPLEMENTARY EXAMINATION PAPER 2018

TITLE OF PAPER

WASTEWATER MANAGEMENT

COURSE CODE

EHM 418

DURATION

2 HOURS

MARKS

100

INSTRUCTIONS

READ THE QUESTIONS & INSTRUCTIONS

CAREFULLY

ANSWER ANY FOUR QUESTIONS

EACH QUESTION <u>CARRIES 25</u> MARKS.

: WRITE NEATLY & CLEARLY

: NO PAPER SHOULD BE BROUGHT INTO THE

EXAMINATION ROOM.

: BEGIN EACH QUESTION ON A SEPARATE

SHEET OF PAPER.

DO NOT OPEN THIS QUESTION PAPER UNTIL PERMISSION IS GRANTED BY THE INVIGILATOR.

QUESTION ONE

1A.	A wastewater is to be treated in a complete mix reactor. Assuming that the
	reaction rate is irreversible and first order (r = -kC) with a reaction rate
	coefficient equal to 0.15 /days, determine the flow rate that can be treated if the
	reactor has a volume of 20 m ³ and 98% treatment efficiency is required.
	[13 Marks]
1B.	What volume would be required to treat the flow rate determined in question 1A
	above if the required treatment efficiency is 92%? [12 Marks]

QUESTION TWO (Each question below carries five marks)

- 2A. List the chemicals that are commonly used for the precipitation of metals from waste waters.
- **2B.** Compare the efficiency of chemical removal of phosphorous in i) primary settlement tanks and ii) secondary settlement tanks.
- **2C.** Describe with the help of a chemical equation how ammonia in wastewater may be removed by chemical oxidation processes.
- **2D.** Compare the energy efficiency of ozone/UV advanced oxidation process with that of hydrogen peroxide/UV processes for the treatment of wastewaters.
- 2E. Discuss possible chemical pretreatment methods for scale control of wastewater that is to be treated with reverse osmosis and nano filtration processes.

QUESTION THREE

- **3B.** Given the chemical reaction equation for the conversion of glucose to cell mass below, compute the cell yield for this reaction.

$$3C_6H_{12}O_6 + 8O_2 + 2NH_3 \rightarrow 2C_5H_7NO_2 + 8CO_2 + 14H_2O$$
 [5 Marks]

3C. The rate of substrate utilization r_{su} for anaerobic processes can be expressed in terms of the well-known Monod equation:

$$r_{su} = \frac{-kSX}{K_s + S}$$

QUESTION FOUR

A sludge collected from primary settlement tank has an initial solids content (on dry basis) of 435 kg. Using the solids processing data given in the table below, determine:

- i. The sludge volume before anaerobic digestion[13 Marks]
- ii. The sludge volume after anaerobic digestion[12 Marks]

Parameter	Primary	Digested
Solids (%) (mass basis)	5	10
Volatile volume (%)	60	60 (destroyed)
Specific gravity of fixed solids	2.5	2.5
Specific gravity of volatile solids	≈ 1.0	≈ 1.0

QUESTION FIVE (Each question below carries 5 marks)

- 5A. List the causes of possible toxicity in activated sludge processes and state the means of reducing or controlling toxicity.
- **5B.** Describe the possible causes of sludge bulking in activated sludge processes.
- 5C. List the four groups of anaerobic bacteria that are involved in the degradation of organic matter in anaerobic process and state in each category the conversion of organic matter that takes place.
- 5D. Discuss the adverse effects of the presence of oxidized sulfur compounds such as sulfate, sulfite and thiosulfate on the performance of anaerobic treatment process of wastewater.
- **5E.** Indicate possible remedial actions for the following operational problems in wastewater treatment:
 - i. Poor BOD removal in trickling filters. (3 marks)
 - ii. Bulking sludge in activated sludge processes.(2 marks)