

UNIVERSITY OF SWAZILAND Faculty of Health Sciences Department of Environmental Health Science

B.Sc. DEGREE IN ENVIRONMENTAL HEALTH SCIENCE

MAIN EXAMINATION PAPER 2017

TITLE OF PAPER

URBAN WATER TREATMENT

COURSE CODE

EHS 222

DURATION

2 HOURS

MARKS

100

INSTRUCTIONS

READ THE QUESTIONS & INSTRUCTIONS

CAREFULLY

ANSWER ANY FOUR QUESTIONS

EACH QUESTION **CARRIES 25** MARKS.

WRITE NEATLY & CLEARLY

NO PAPER SHOULD BE BROUGHT INTO THE

EXAMINATION ROOM.

BEGIN EACH QUESTION ON A SEPARATE SHEET

OF PAPER.

DO NOT OPEN THIS QUESTION PAPER UNTIL PERMISSION IS GRANTED BY THE INVIGILATOR.

QUESTION ONE (5 Marks each)

1A. Multiple barrier system is a term that is commonly used in water treatment. Explain what this term means and state its importance in modern water treatment processes.

.....[5 marks]

1B. List and describe the factors that influence the location of intakes to water treatment plants.

..... [5 marks]

1C. Figure Q1-1 below shows a multiple intake reservoir system. Describe the advantages of this multiple intake system and the reason for providing a siphon

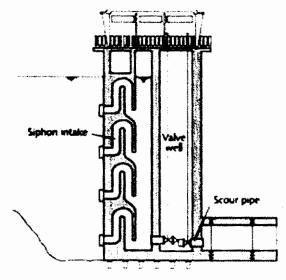


Figure Q1-1 Multiple level intake system with siphon intake



Figure Q1-2 A structure (appearing white) provided inside a water body

QUESTION TWO

- 2A. Discuss the appropriateness of plain sedimentation tanks for the following cases.
 - i. The raw water contains mostly sand and silt.[1 mark]
 - ii. The raw water contains mostly colloidal suspensions....[1 mark]
 - iii. The raw water contains mostly a mixture of sand, silt and colloidal suspensions.[1 mark]
 - iv. The raw water contains largely suspended algal material....[1 mark]
 - v. The raw water has passed through infiltration gallery.[1 mark]
- **2B.** Looking at the stability diagram of copper metal shown in Figure Q2-1 below, discuss in detail the potential for corrosion of copper. [5 marks]

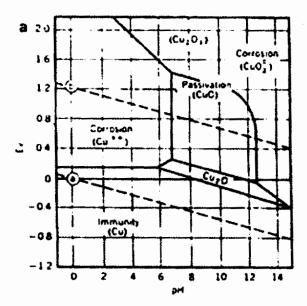


Figure Q2-1 Stability diagram of copper metal

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	i.	The pH of the water decreased[1 mark]
	ii.	The alkalinity of the water increased[1 mark]
	iii.	The calcium content of the water increased[1 mark]
	iv.	The temperature of water increased [1 mark]
	ν.	The sodium content of the water increased[1 mark]

QUESTION THREE

- 3A. Define the following terms: i) suspension ii) colloid iii) floc iv) coagulation v) flocculation. [5 marks]
- 3C. Describe i) Ortho-kinetic flocculation ii) Peri-kinetic flocculation. Indicate the instances in which each of these types of flocculation arise in water treatment. Indicate the factors that affect the extent of flocculation for each type. ...[5 marks]

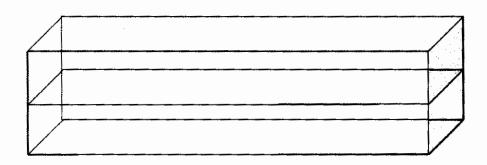


Figure Q3-1: Rectangular sedimentation tank in which extra bottom is provided in the middle of the tank.

QUESTION FOUR

- **4D.** Figure Q4-1 below shows the cross section of different layers existing in slow sand filter just before the time for re-sanding is reached. Draw the cross section of the different layers in the slow sand filter immediately after sanding.[5 marks]

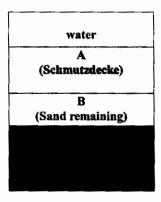


Figure Q4-1 Layers of slow sand filter just before re-sanding

4E. List possible operational faults of slow sand filter. [5 marks]

QUESTION FIVE

5A. Describe how disinfection byproducts are formed and the measures that can be taken to minimize the formation of disinfection byproducts in water treatment.

.....[5 marks]

- **5B.** Discuss the advantages and limitations of the following two disinfectants when used in water treatment:
 - i. Ultra violet ray[2 marks]
 - ii. Chloramine[3 marks]

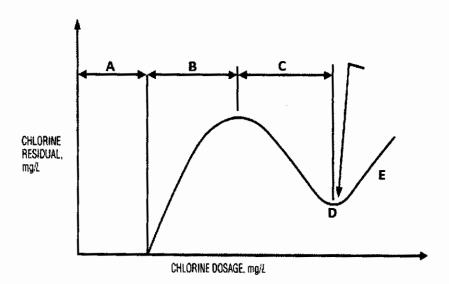


Figure Q5-1 Chlorine residual curve

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5D.	You are required to design a chlorine dosing system using calcium hypochlorite as a disinfectant. The water to be disinfected has passed through a filtration system and is led by gravity into to a chlorine tank. Outline with the help of a sketch how you would achieve this design
5E.	Explain the variation of the effectiveness of chlorine disinfection with water temperature in terms of activation energy

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