

UNIVERSITY OF SWAZILAND Faculty of Health Sciences Department of Environmental Health Science

B.Sc. DEGREE IN: - ENVIRONMENTAL MANAGEMENT &
OCCUPATIONAL SAFETY AND HEALTH
- ENVIRONMENTAL MANAGEMENT
AND WATER RESOURCES

MAIN EXAMINATION PAPER 2017

TITLE OF PAPER

WASTEWATER MANAGEMENT

COURSE CODE

EHM 418

DURATION

2 HOURS

MARKS

100

INSTRUCTIONS

READ THE QUESTIONS & INSTRUCTIONS

CAREFULLY

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ANSWER ANY FOUR QUESTIONS

EACH QUESTION **CARRIES 25** MARKS.

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WRITE NEATLY & CLEARLY

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NO PAPER SHOULD BE BROUGHT INTO THE

EXAMINATION ROOM.

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BEGIN EACH QUESTION ON A SEPARATE

SHEET OF PAPER.

DO NOT OPEN THIS QUESTION PAPER UNTIL PERMISSION IS GRANTED BY THE INVIGILATOR.

1A.	Show the out	et tracer test profile diagram using salt tracer for the following flow
	conditions:	
	i	. Plug flow tank [2 ½ marks]
	ii	. Completely mixed flow tank [2 ½ marks]

Assume that 100 gm of salt is added to the inlet and the volume of the tank in each case is 4 m³. The wastewater flow rate is 2 m³/hr.

- - i. Lagoons
 - ii. Aerated lagoons

QUESTION ONE (Marks are indicated for each question)

- iii. Activated sludge.

QUESTION TWO (Each question below carries 5 marks)

2A.	Describe the role of surfactants in the formation of stable oil emulsions in
	wastewater.

- **2B.** What test can be used to determine the concentration of settleable solids in wastewater?
- **2C.** What treatment techniques can be used for the removal of volatile organic carbon from wastewater?
- **2D.** For what type of situations are mechanically cleaned bar screens preferred to manually cleaned bar screens?
- **2E.** List the types of oil that may be present in wastewater treatment processes and the treatments needed to remove each type of oil.

QUESTION THREE (Each question below carries 5 marks)

- 3A. Define the following terms used in relation to chemical unit processes:
 - i. Extinction coefficient [1 mark]
 - ii. Gas stripping ...[1 mark]
 - iii. Advanced oxidation[1 mark]
 - iv. Photolysis[1 mark]
 - v. Reverse osmosis[1 mark]
- **3B.** Figure Q3-1 below shows the solubility diagrams of aluminum and iron with respect to pH of wastewater. Discuss the relative importance of pH control on the efficiency of coagulation in wastewater using these two different metals.

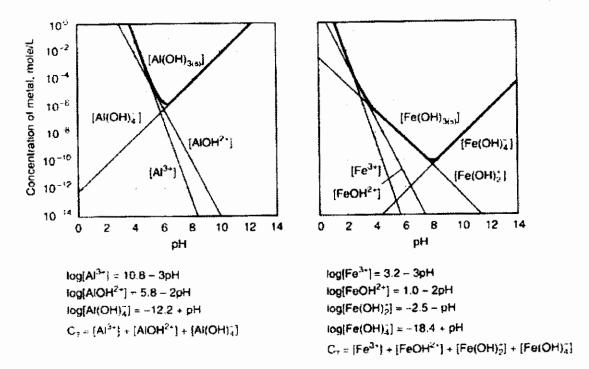


Figure Q3-1: Solubility diagrams of aluminum and iron with respect to pH of wastewater

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3C.	List possible chemical treatment methods employed for the removal of phosphorus
	from wastewater

- **3D.** List five typical applications of chemical oxidation processes in wastewater treatment
- **3E.** List the advantages and disadvantages of advanced oxidation processes for wastewater treatment.

QUESTION THREE (5 marks for each question)

4A. The chemical formula shown below represents the breakdown of glucose by aerobic bacteria and the formation of new bacterial cells and metabolic byproducts. Based on this reaction, determine the cell yield coefficient

$$3C_6H_{12}O_6 + 8O_2 + 2NH_3 \rightarrow 2C_5H_7NO_2 + 8CO_2 + 14H_2O$$

- **4B.** Describe with the help of a diagram the pre anoxic and post anoxic denitrification processes.
- **4C.** Describe the sources of sludge bulking and the mechanisms employed to control sludge bulking in activated sludge processes
- **4D.** Describe the possible causes and remedies for clogging in percolating filter systems
- **4E.** List possible advantages and disadvantages of anaerobic treatment processes

QUESTION FIVE (Each question below carries 5 marks)

- **5A.** Describe the sources of solids in wastewater treatment processes and the sequential treatment/stabilization processes typically employed to produces bio-solids suitable for agricultural application.
- **5B.** List possible sludge processing strategies used to reduce or eliminate pathogens from sludge.
- 5C. Describe briefly the four discrete steps used in environmental risk analysis
- 5D. Describe water pinch analysis and the benefits it provides.
- **5E.** Describe the sources of odours in wastewater treatment plants and the mechanism that can be placed to minimize odours.