# UNIVERSITY OF SWAZILAND Faculty of Health Sciences Department of Environmental Health Sciences

### **BSc Environmental Health**

## **MAIN EXAMINATION PAPER DECEMBER2014**

TITLE OF PAPER :

INDUSTRIAL WASTE MANAGEMENT I

COURSE CODE

EHS:553

DURATION

: 2 HOURS

MARKS

100

INSTRUCTIONS

THERE ARE FIVE QUESTIONS IN THIS EXAM

: ANSWER ANY 4 OF THE 5 QUESTIONS

: EACH QUESTION CARRIES 25 MARKS

: NO PAPER SHOULD BE BROUGHT INTO OR OUT OF THE

**EXAMINATION ROOM** 

EHS 553 DECEMBER 2014

#### **QUESTION ONE (25 Marks)**

A wastewater contains the following: 130 mg/L ethylene glycol 85 mg phenol 30 mg/L sulfide (S<sup>2</sup>) 100 mg/L ethylene diamine hydrate (ethylenediamine is essentially non-biodegradable). 1A.Compute the COD and TOC ......[10 marks] 1C.After treatment the BOD<sub>5</sub> is 25 mg/L. Estimate the COD of the effluent (  $k_{10} = 0.1/day$ ).....[7 marks] Chemical equations of oxidation Ethylene glycol:  $C_2H_6O_2 + 2.5 O_2 \rightarrow 2CO_2 + 3H_2O$ Phenol: ii)  $C_6H_6O + 7O_2 \rightarrow 6CO_2 + 3H_2O$ Ethylene diamine hydrate:  $C_2H_{10}N_2O + 2.5 O_2 \rightarrow 2CO_2 + 2H_2O + 2NH_3$ iv) Sulfide:

 $S^{2-} + 2O_2 \rightarrow SO_4^{2-}$ 

#### QUESTION TWO (25 Marks) (Note each question carries 5 marks)

- 2A. What are the main factors that must be considered when designing a pilot scale experiment for industrial waste treatment?
- 2B. What is the effect of temperature inversion with respect to emissions of pollutants from industries? When is temperature inversion likely to occur?
- **2C.** Give four examples of bio refractory organics.
- **2D.** Compare the variation of the re-aeration coefficient and de-aeration coefficient with water temperature and the implication on thermal pollution of water sources from industrial discharges.
- 2E. Discuss the type of sampling appropriate for each of the following parameters:
  - i. pH
  - ii. BOD for the design of activated sludge process unit
  - iii. BOD for the design of aerated lagoons
  - iv. Nutrients for the design of sequencing batch reactors.

#### **QUESTION THREE** (25 Marks)

An industry discharges waste water in to an equalization basin between 0hr and 16 hr. Between 0hr and 8hr the total amount of waste water discharged was 12 Million liters, discharged uniformly in to the equalization basin. Between 8hr and 16 hr, the total amount of waste water discharged in this period was 16 Million liters discharged at a uniform rate. Between 16hr and 24hr the industry is closed so that there is no waste water flow in to the equalization basin. If the outflow from the equalization basin is such that 30 % of the total out flow occurs during the day (between 0 hour and 12 hour) at a uniform rate and 70 % during the night (between 12 hr and 24 hour) at uniform rate,

3A.Determine the volume of the equalization basin.	13 Marks]
3B. Determine the volume of the equalization basin if the outflow from t	he equalization tank
was uniform throughout the 24 hour period and compare with the re-	esult obtained in 3A
above.	[ 12 marks]

QUESTION FOUR (2:	Marks)
-------------------	--------

4A.	What is the imp	ortance of grit chambers in waste water treatment?[4 Marks ]
4B.		use of short circuiting in sedimentation tanks and the methods to prevent[4 Marks]
4C.		tween the following oil suspension in wastewater and indicate the methods
	for their remova	ll from wastewater:
		i) Free oil
		ii) Physical emulsion
		iii) Chemical emulsion
	•	iv) Dissolved oil[7 Marks ]
4D.	·	y the nature of wastes generated and treatment alternatives for wastewater
	generated in pr	armaceutical industries [ 10 Marks]

## QUESTION FIVE (25 Marks)

<b>5A.</b>	wastewaters		
	wase waters.		
5B.	Describe treatment mechanisms for:		
	i) Laundry wastes containing synthetic detergents[3 marks]		
	ii) Polymer wastes from latex manufacture		
5C.	List the technologies available for heavy metal removal from industrial wastewaters		
	[ 5 Marks]		
5D.	Describe the treatment method that can be used for the removal of cyanide form		
	industrial wastewaters [ 5 Marks]		
5E.	Describe the advantages and disadvantages of employing wet air oxidation processes for		
	the treatment of industrial wastewaters. [5Marks]		