

### **FACULTY OF HEALTH SCIENCES**

#### **B.Sc. ENVIRONMENTAL HEALTH SCIENCE**

## MAIN EXAMINATION:

TITLE OF PAPER:

**FOOD ANALYSIS** 

COURSE CODE:

EHM325

**DURATION:** 

2 HOURS

DATE:

MAY 2015

**INSTRUCTIONS:** 

- 1. READ THE QUESTIONS CAREFULLY.
- 2. ANSWER ANY 4 OUT OF 5 QUESTIONS.
- EACH QUESTION CARRIES 25 MARKS. WHERE A QUESTION IS SUBDIVIDED INTO PARTS, THE MARK FOR EACH PART IS SHOWN IN BRACKETS.
- 4. NO PAPER SHOULD BE BROUGHT INTO THE EXAMINATION ROOM.
- 5. BEGIN EACH QUESTION ON A SEPARATE SHEET OF PAPER.

**SPECIAL REQUIREMENTS: NONE** 

DO NOT OPEN THE QUESTION PAPER UNTIL INSTRUCTED TO DO SO BY THE INVIGILATOR.

### **QUESTION 1**

- a. Showing reactions involved where applicable, explain the role of each of the following reagents in the Kjeldahl method of protein determination:
  - i. Sulphuric acid. [3]
  - ii. Sodium sulphate. [2]
  - iii. Copper sulphate. [2]
  - iv. Sodium hydroxide. [3]
- b. Outline the advantages and disadvantages of each of the following protein determination methods:
  - i. Biuret method. [7]
  - ii. Dye binding method. [8]

[25]

# **QUESTION 2**

Write notes on the following:

- a. Partition coefficient [5]
- b. Gradient elution [5]
- c. Temperature programming. [5]
- d. Column efficiency. [10]

[25]

## **QUESTION 3**

- a. Briefly explain how moisture is detected in the volumetric determination of moisture by Karl Fischer method. [10]
- Using diagrams to illustrate, and stating examples and the solvents involved, discuss the principles in moisture determination in food using reflux distillation techniques.

[25]

## **QUESTION 4**

- a. Compare and contrast the following:
  - i. Dry ashing and wet oxidation. [5]
  - ii. Water activity and moisture content. [5]
  - iii. Crude fibre and dietary fibre. [5]
  - iv. Iodine value and peroxide value. [10]

[25]

# Question 5

- a. Explain how analytes are detected during thin layer chromatography. [5]
- b. Explain how column efficiency is characterized under the following headings:
  - i. Resolution. [5]
  - ii. Height equivalent to theoretical plates. [10]
- c. Explain the principle responsible for separation of solutes in Gas Chromatography. [5]

[25]

#### **END OF EXAMINATION**