



2ND SEM. 2016/17

**PAGE 1 OF 4
FNS 102 (M)**

UNIVERSITY OF SWAZILAND

FINAL EXAMINATION PAPER

- PROGRAMME** : **FOOD SCIENCE, NUTRITION AND TECHNOLOGY YEAR II**
- COURSE CODE** : **FNS 102**
- TITLE OF PAPER** : **FOOD CHEMISTRY**
- TIME ALLOWED** : **TWO (2) HOURS**
- INSTRUCTIONS** : **ANSWER QUESTION ONE (1) AND ANY OTHER TWO (2) QUESTIONS.**

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QUESTION 1 (COMPULSORY)

- (a) Describe the following food dispersions and give **one (1)** food example in each case.
- A solution
 - Gel
 - Colloidal suspension

(6 Marks)

- (b) Draw the electronic configuration of the oxygen atom and show how the Sp^3 hybrid orbitals are formed and also draw the water molecules and explain why the bond angle between the hydrogen atoms is 104.5° .

(9 Marks)

- (c) Draw the structure of linolenic acid (6, 9, 12-Octadecatrienoic acid). What is the omega name of this fatty acid?

(7 Marks)

- (d) Draw the structure of the following sugars and name the glycosidic bond in the disaccharide sugars.
- α -D-Glucose
 - β -D-Galactose
 - Maltose
 - Cellobiose

(14 Marks)

- (e) Explain the following:-

- Why do long chain fatty acids have higher melting points than short chain fatty acids?
- Why do the melting points of fatty acids decrease with increase in the number of double bonds?

(4 Marks)**[TOTAL MARKS = 40]**

QUESTION 2

- (a) Draw the 3 dimensional network structure of water and explain why water has low viscosity.

(6 Marks)

- (b) Show how lipase can cleave all three fatty acids from a triglyceride. Explain **two (2)** disadvantages and **one (1)** advantage of this enzyme in food.

(9 Marks)

- (c) Describe the role of fat in the following baked products:-

- i. Pastry
- ii. Cake
- iii. Bread

(9 Marks)

- (d) Name **three (3)** groups of naturally occurring colour compounds in plant foods.

(6 Marks)

[TOTAL MARKS = 30]

QUESTION 3

- (a) Explain the following protein structures:-

- i. Primary structure
- ii. Secondary structure

(4 Marks)

- (b) Describe the effect of the following agents on protein denaturation:-

- i. Change in temperature
- ii. Hydrostatic pressure
- iii. pH

(9 Marks)

- (c) With the aid of a diagram explain the composition of starch stating the monosaccharide sugar units and type of bonds between the units.

(12 Marks)

- (d) What are the effects of the presence of the following minerals in food?

- i. Copper and iron in fatty foods
- ii. Calcium in cheese

(5 Marks)

[TOTAL MARKS = 30]

QUESTION 4

- (a) What is an emulsion? Explain the two (2) types of emulsions giving an example in each case.

(8 Marks)

- (b) What is the Maillard browning reaction and list **three (3)** conditions that favours the reaction?

(9 Marks)

- (c) Explain the action of the following enzymes:-

- i. α -Amylase on starch
- ii. β -Amylase on starch
- iii. Glucose isomerase on glucose

(9 Marks)

- (d) Oxidation of D-glucose results in the production of D-glucono- δ -lactone. What is the food application of D-glucono- δ -lactone.

(4 Marks)

[TOTAL MARKS = 30]