

2<sup>ND</sup> SEM. 2015/16

PAGE 1 OF 5

## UNIVERSITY OF SWAZILAND

## FINAL EXAMINATION PAPER

**PROGRAMME** 

FOOD SCIENCE, NUTRITION AND

TECHNOLOGY YEAR III

**COURSE CODE** 

**FSNT 307** 

:

:

TITLE OF PAPER

FOOD NUTRIENT ANALYSIS

TIME ALLOWED

TWO (2) HOURS

**INSTRUCTIONS** 

ANSWER QUESTION ONE (1)

AND ANY OTHER TWO (2)

QUESTIONS. STATISTICAL TABLES AND FORMULA ARE PROVIDED AT THE END OF THE QUESTION PAPER

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### QUESTION 1 (COMPULSORY)

(a) Explain the following terms:-

	and termine.	
1.	Sampling	
ii	Finite population	(2 Marks)
:::	Three population	(2 Marks)
111.	Heterogeneous population	
		(2 Marks)

- (b) Discuss four (4) changes that could occur to the food sample and how these changes can be slowed down or stopped. (20 Marks)
- (c) What is the function of sulphuric acid and amyl alcohol in the Geber method for fat content determination in milk? (2 Marks)
- (d) The equation y = 0.1933x + 0.0134 was obtained for the standard calibration curve (x-axis = concentration and y-axis = absorbance) for vitamin A.
  - i. Using the equation, calculate the concentration of each of the following samples. (4 Marks)

Sample	Abs
A	0.35
B	0.38
C	0.37
D	0.34

- ii. Calculate the mean of the samples. (2 Marks)
- iii. Calculate the standard deviation of the samples. (4 Marks)
- iv. Calculate the coefficient of variation. (2 Marks)

[TOTAL MARKS = 40]

### **QUESTION 2**

- (a) Discuss the three (3) forms of water in food and indicate which form of water is not determined by the oven drying method. (9 Marks)
- (b) Explain the principles behind the following methods for moisture content determination in food:-

ii.	Microwave drying method Distillation method Gas production method	(2 Marks) (5 Marks)
		(4 Marks)

(c) Explain the digestion, distillation and titration steps in the Kjeldahl protein determination method. (10 Marks)

[TOTAL MARKS = 30]

#### **QUESTION 3**

(a) Describe the following types of error errors will affect the results:-	rs in nutrient analysis and explain how these
i Determinate arror	

i.	Determinate error	
ii.	Indeterminate error	(3 Marks)
iii.	Gross error	(3 Marks)
	0.000 01101	(3 Marks)

(b) Explain the following functions of HPLC:-

i.	Chemical separation	
ii.	Purification	(3 Marks)
iii.	Identification	(3 Marks)
iv.	Quantification	(3 Marks)
	Quantification	(4 Marks)

(d) Describe the Soxhlet extraction method for crude fat determination. (8 Marks)

[TOTAL MARKS = 30]

#### **QUESTION 4**

(a) The following data were obtained for the sugar concentration of a sugar solution: 4.85, 6.18, 6.28, 6.49, 6.69. Should 4.85 be rejected or retained at the following confidence levels:
(10 marks)

i. 95% ii. 90%

(b) Explain the principles in crude fibre determination method. (8 Marks)

(c) Briefly explain the use and principle of atomic absorption spectroscopy.

(d) What causes fluorescence in some compounds such as aflatoxins under UV light?

(e) Describe the equipment in gas chromatography (GC). (2 Marks) (6 Marks)

[TOTAL MARKS = 30]

Formula Mean

$$\bar{X} = \frac{\sum x}{n}$$

Standard deviation

$$S = \sqrt{\frac{\sum (x - \bar{x})^2}{n - 1}}$$

Coefficient of variation (CoV)

$$CoV = \frac{S}{\bar{X}} \times 100$$

Equation of a straight line

Slope

$$y = mx + c$$

$$m = \frac{\sum xy}{\sum x^2}$$

y-Intercept

$$c = \bar{y} - m\bar{x}$$

Correlation coefficient

$$r = \frac{\sum xy}{\sqrt{(\sum x^2)(\sum y^2)}}$$

Outlier

$$Q \ value = \frac{x_2 - x_1}{W}$$

Where  $x_1 = Questionable value$ 

 $x_2 =$ Closest value to X1

W = Range (Highest value - lowest value)

# Critical values for Dixon's Q-test

	Q.,,,	Q	Q <sub>m</sub>
	CL at 90%	CL at 95%	CL at 99%
3	0.941	0.970	0.994
4	0.765	0.829	0.926
5	0.642	0.710	0.821
6	0.560	0.625	0.740
7	0.507	0.568	0.680
8	0.468	0.526	0.634
9	0.437	0.493	0.598
10	0.412	0.466	0.568

The data is discarded if the calculated Q-value is higher than the tabulated value Q-critical