

2ND SEM. 2013/14

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UNIVERSITY OF SWAZILAND

FINAL EXAMINATION PAPER

PROGRAMME

: FOOD SCIENCE, NUTRITION AND TECHNOLOGY

YEAR II

COURSE CODE

: FSNT 206

TITLE OF PAPER

: FOOD CHEMISTRY

TIME ALLOWED

: TWO (2) HOURS

INSTRUCTIONS

: ANSWER QUESTION ONE (1) AND ANY OTHER

TWO (2) QUESTIONS.

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QUESTION 1 (COMPULSORY)

- (a) Describe the following food dispersions and give one (1) food example in each case:
 - i. Water-in-oil emulsion
 - ii. Gel
 - iii. A solution

(6 Marks)

(b) Draw a water molecule showing the sp3 hybridized orbitals and explain why the bond angle between the hydrogen atoms in the water molecule (H₂O) is 104.5°.

(5 Marks)

(c) Draw the structure of linolenic acid (6, 9, 12-Octadecatrienoic acid). What is the omega name of this fatty acid?

(7 Marks)

(d) Draw the structures of the following sugar molecules and name the glycosidic bond in the disaccharide sugars:-

i. α-D-Glucose

(3 marks)

ii. β-D-Galactose

(3 marks)

iii. Maltose

(4 marks)

iv. Cellobiose

(4 marks)

v. Sucrose

(4 marks)

(18 Marks)

- (e) Explain the following:
 - i. Why long chain fatty acids have higher melting points than short chain fatty acids?
 - ii. Why the melting points of fatty acids decreases with increase in the number of double bonds?

(4 Marks)

[TOTAL MARKS = 40]

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QUESTION 2

(a) Draw the 3 dimensional network structure of water and explain why water has low viscosity.

(6 Marks)

(b) Show how lipase can cleave all three fatty acids from a triglyceride. Explain two (2) disadvantages and one (1) advantage of this enzyme in food.

(9 Marks)

- (c) Describe the role of fat in the following baked products:
 - i. Pastry
 - ii. Cake
 - iii. Bread

(9 Marks)

(d) Name three (3) groups of naturally occurring colour compounds in plant foods.

(6 Marks)

[TOTAL MARKS =30]

QUESTION 3

- (a) Explain the following protein structures:
 - i. Primary structure
 - ii. Secondary structure

(4 Marks)

- (b) Describe the effect of the following agents on protein denaturation:
 - i. Change in temperature
 - ii. Hydrostatic pressure
 - iii. pH

(9 Marks)

(c) With the aid of a diagram explain the composition of starch stating the monosaccharide units and type of bonds between the units.

(12 Marks)

- (d) What are the effects of the presence of the following minerals in food?
 - i. Copper and iron in fatty foods
 - ii. Calcium in cheese

(5 Marks)

[TOTAL MARKS = 30]

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QUESTION 4

(a) Describe the structure, properties and uses of cyclodextrins.

(7 Marks)

(b) What is Maillard browning reaction and list **three** (3) conditions that favour the reaction?

(9 Marks)

- (c) Explain the action of the following enzymes:
 - i. α -Amylase on starch
 - ii. β-Amylase on starch
 - iii. Glucose isomerase on glucose

(9 Marks)

(d) Oxidation of D-glucose results in the production of D-glucono- δ -lactone. What is the food application of D-glucono- δ -lactone?

(5 Marks)

[TOTAL MARKS = 30]