

2ND SEM. 2011/2012

UNIVERSITY OF SWAZILAND

FINAL EXAMINATION PAPER

PROGRAMME

BACHELOR OF SCIENCE IN FOOD

SCIENCE, NUTRITION & TECHNOLOGY

YEAR II

COURSE CODE

FSNT 206

TITLE OF PAPER

FOOD CHEMISTRY

TIME ALLOWED

TWO (2) HOURS

INSTRUCTIONS

ANSWER QUESTION ONE (1)

AND ANY OTHER (2) QUESTIONS

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QUESTION 1 [COMPULSORY]

a. Define primary and tertiary structure of protein.

[8 marks]

b. Differentiate between "iodine value" and "saponification value".

[8 marks]

c. What is the role of an emulsifier in food systems?

[8 marks]

d. Milk was heated at 120° C for 10 minutes and it turned brownish. Give two possible reactions that could have caused the browning colouration.

[8 marks]

e. Discuss the role of hydrogen bonding in aqueous solutions.

[8 marks]

[TOTAL MARKS = 40]

OUESTION 2

a) Discuss the conditions that favour Maillard reactions and give at least three negative effects of the reactions.

[20 Marks]

b) Starch can be digested by alpha amylase while cellulose is not digestible. What is the cause of the difference in digestibility?

(10 Marks)

[TOTAL MARKS = 30]

QUESTION 3

a) Discuss the importance of fat in the human diet.

[15 Marks]

b) What do you understand by the term essential fatty acid?

[6 Marks]

c) Define hydrogenation as it refers to oil and give one advantage and two disadvantages of the process.

[9 marks]

[TOTAL MARKS = 30]

QUESTION 4

a. Discuss the major causes of post harvest deterioration in fruits and vegetables.

[12 Marks]

b. When a potato is cut, the cut surface is likely to turn brown. What is the difference between this browning and the browning observed in fried potato chips?

[8 Marks]

c. Pectin is a good gelling agent. Explain how this happens and its application in food processing.

(10 marks)

[TOTAL MARKS = 30]