

# 1<sup>ST</sup> SEM. 2007/2008

## UNIVERSITY OF SWAZILAND

#### FINAL EXAMINATION PAPER

**PROGRAMME:** 

**BACHELOR OF SCIENCE IN** 

AGRONOMY; BACHELOR OF

SCIENCE IN ANIMAL SCIENCE; BACHELOR OF SCIENCE IN FOOD

SCIENCE, NUTRITION AND

TECHNOLOGY; AND BACHELOR OF SCIENCE IN HORTICULTURE

YEAR II

**COURSE CODE:** 

APH 203

TITLE OF PAPER:

**BIOCHEMISTRY** 

TIME ALLOWED:

TWO (2) HOURS

**INSTRUCTIONS:** 

**ANSWER ANY 4 QUESTIONS** 

THIS PAPER MAY NOT BE OPENED UNTIL THE CHIEF INVIGILATOR HAS GRANTED PERMISSION.

## **QUESTION 1**

Describe and illustrate the following:

a) Eicosanoids (14 Marks)

b) Hydrogen bonding of water molecules (11 Marks)

## **QUESTION 2**

Using structures to illustrate your answers explain the major differences between the following:

a) Glycosidic bond and peptide bonds

(8 Marks)

b) Reducing and non reducing sugars

(9 Marks)

c) Nucleoside and Nucleotide (8 Marks)

#### **QUESTION 3**

Study the structure of the biomolecule shown below and answer the following questions:

a)	Identify the biomolecule	(1 Marks)
b)	Illustrate and identify the tautomer of the biomolecule	(4 Marks)
c)	Illustrate and identify the amino sugar of the biomolecule	(4 Marks)
d)	Illustrate and identify the alditol of the biomolecule	(4 Marks)
e)	Illustrate and identify the anomeric pair of the biomolecule	(6 Marks)
f)	Illustrate and identify the disaccharide made by the	
	biomolecule and its tautomer.	(6 Marks)

## **QUESTION 4**

a). Briefly discuss three advantages of using enzymes as biological catalysts. (9 Marks)

- b). Discuss how the following may affect enzymic reaction rate:
  - i) temperature
  - ii) pH
  - iii) Enzyme concentration
  - iv) Substrate concentration · (16 Marks)

# **QUESTION 5**

Using structures to support your answer, briefly describe two examples of:

a)	Unsaturated fatty acids	(5 Marks)
b)	Sterol lipids	(5 Marks)
c)	Non protein amino acids	(5 Marks)
d)	Structural polysaccharides	(5 Marks)
e)	Water soluble vitamins	(5 Marks)